



Application Note AN-S-373

Chloride in potassium bicarbonate and potassium chloride effervescent tablets for oral solution

Method validation according to the U.S. Pharmacopoeia

Potassium chloride and potassium bicarbonate effervescent tablets are used to prevent low levels of potassium in blood [1]. Using the monographs from the United States Pharmacopeia and National Formulary (USP-NF) allows pharmaceutical manufacturers and labs to fulfill strict quality regulations for drugs and formulations.

The USP has embarked on a global initiative to modernize many existing monographs. The monograph «Potassium Bicarbonate and Potassium Chloride Effervescent Tablets for Oral

Solution» comprises different methods to determine potassium, sodium, but also chloride in these tablets [2]. Ion chromatography (IC) with suppressed conductivity detection has been approved by the USP as a validated method to quantify chloride content in potassium bicarbonate and potassium chloride effervescent tablets for oral suspension [2]. The Metrosep A Supp 16 - 100/4.0 (L91) column provides the required separation of chloride. The method is validated according to USP General Chapter <621> Chromatography, system suitability [3].

SAMPLE AND SAMPLE PREPARATION

Sample analyses are performed with a solution of the respective effervescent tablets. No

additional sample preparation is required.



Figure 1. Instrumental setup including a 930 Compact IC Flex Oven/SeS/PP and an 858 Professional Sample Processor.

A sample stock solution (nominally 4434.52 µg/mL chloride) is prepared by adding 50 g (equivalent to 10 tablets weight) of finely powdered potassium bicarbonate and potassium chloride effervescent tablets for oral solution to a 2000 mL volumetric flask. The powder is dissolved in 200 mL ultrapure water (UPW). After effervescence ceases, the volumetric flask is filled up to the mark. A small volume (1.692 mL) of this stock solution is transferred to a 500 mL volumetric flask and filled up to the mark with UPW. This final sample solution nominally contains 15.0 µg/mL chloride.

The working standard solution of 15 µg/mL is prepared from a USP Potassium Chloride RS

standard.

Samples and standard solutions are injected directly into the IC using an 858 Professional Sample Processor (**Figure 1**). Separation of chloride from other anions is performed using a Metrosep A Supp 16 - 100/4.0 column. This anion-exchange column, consisting of a strong ion exchanger made from monodisperse porous polystyrene/divinyl benzene beads combined with quaternary amines, qualifies for certain USP methods using the USP chromatographic column packing L91.

The calibration is performed with a 6-point linear calibration curve using a concentration range of 2.25–22.50 µg/mL chloride. The sample is then analyzed in duplicate.

Table 1. Requirements for the IC method for chloride determination as per USP Monograph «Potassium Bicarbonate and Potassium Chloride Effervescent Tablets for Oral Solution» [2].

Column with L91 packing	Metrosep A Supp 16 - 100/4.0
Eluent	15 mmol/L sodium carbonate, 1.5 mmol/L sodium hydroxide
Flow rate	0.8 mL/min
Temperature	45 ° C
Injection volume	20 µ L
Detection	Suppressed conductivity

RESULTS

The IC assay for chloride content was validated according to the USP Monograph «Potassium Bicarbonate and Potassium Chloride Effervescent Tablets for Oral Solution» [2]. The accuracy of the chloride determination in the sample was calculated as 101.2% (**Figure 2**) and

falls into the acceptance criteria. All analytical quality requirements were fulfilled, e.g., the correlation coefficient for chloride was 0.9998, and the relative standard deviation of repeated standard solutions was 0.05% (n = 6) (**Table 2**).

Table 2. Analytical quality criteria for method acceptance according to USP Monograph «Potassium Bicarbonate and Potassium Chloride Effervescent Tablets for Oral Solution» [2].

Parameter	Actual	USP requirement	Status
% RSD	0.05	NMT 0.5	Pass
Tailing factor	1.27	NMT 2.0	Pass
Recovery	101.2%	90–110%	Pass
Resolution	2.48	NLT 1.5	Pass

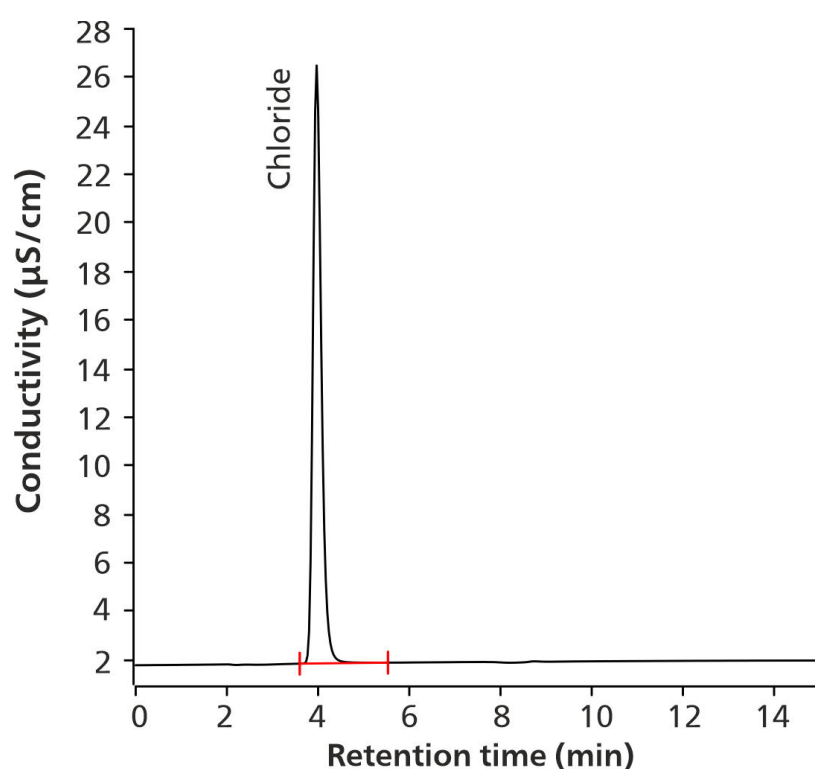


Figure 2. Chromatogram of 15.0 $\mu\text{g/mL}$ chloride in the sample solution (101.1% recovery of the nominal concentration).

The presented IC method for chloride in potassium bicarbonate and potassium chloride effervescent tablets for oral solution is officially included into the USP [2]. Chloride separation is performed with a strong anion-exchanger – the Metrosep A Supp 16 - 100/4.0 column, corresponding to packing material

L91. Robustness and reliability of the method was demonstrated following the guidelines of the USP General Chapter <621> [3]. The presented setup is suitable to quantify chloride according to the USP requirements. Further USP methods are summarized in the flyer «Bring your USP methods up to date!» [4].

REFERENCES

1. Kardalas, E.; Paschou, S. A.; Anagnostis, P.; et al. Hypokalemia: A Clinical Update. *Endocr Connect* **2018**, 7 (4), R135–R146. <https://doi.org/10.1530/EC-18-0109>.
2. *Potassium Bicarbonate and Potassium Chloride Effervescent Tablets for Oral Solution*; Monograph; U.S. Pharmacopeia/National Formulary: Rockville, MD. https://doi.org/10.31003/USPNF_M67253_02_01.
3. <621> *Chromatography, General Chapter*, U.S. Pharmacopeia/National Formulary: Rockville, MD. <https://www.uspnf.com/notices-gc-621-nitr-20220826>.
4. Metrohm AG. Bring Your USP Methods up to Date!, 2023. [8.000.5436EN](https://www.metrohm.com/8.000.5436EN)

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CONFIGURATION



Metrosep A Supp 16 - 100/4.0

Metrosep A Supp 16 ,/,
(EPA 326,DIN EN ISO 11206) Metrosep A Supp 16
- 100/4.0



Metrosep A Supp 16 Guard/4.0

Metrosep A Supp 16 Guard/4.0 保柱可有效保护分析分柱 Metrosep A Supp 16,使其免受污染。保柱采用 «On Column Guard System» 技术,其特征是操作简便。只要将其到分析柱上即可。不需要任何工具。



930 Compact IC Flex Oven/SeS/PP/Deg

930 Compact IC Flex Oven/SeS/PP/Deg 是智能型 Compact 子色器,有柱加炉、**序列抑制**和**蠕**用于抑制器再生,以及内置的**脱气装置**。器可使用各分和方法。典型的用范:

- 子或子定,序列抑制法及



858 Professional Sample Processor

858 Professional Sample Processor 可理体在500 μ L 至 500 mL 之的品。行品移,既可以使用 850 Professional IC System 上的蠕、也可通 800 Dosino 来行。



USP

本宣中 USP 近期完成的色更新行了概述,其在已定了用于多 API、和形物分析的子色法。