Application Note AN-V-197

Indirect determination of iodide in brine with stripping voltammetry

Quantification of iodide in the chlor-alkali process through iodate formation with the hanging mercury drop electrode

Monitoring the iodide concentration in NaCl brine is crucial during membrane process-based chlor-alkali electrolysis. Iodide can easily oxidize to iodate during electrolysis, leading to its precipitation and fouling of the membrane surface. Fouling can reduce the high efficiency of the membrane process and lead to increased energy consumption and decreased product quality. Therefore, monitoring the iodide concentration can help prevent fouling and protect the expensive membranes used in this process.

Stripping voltammetry, with its low detection limit and quick analysis capabilities, emerges as an attractive tool for the analysis of iodide in highly concentrated brines. By utilizing voltammetry, chlor-alkali plants can effectively monitor and manage iodide levels, thus preventing membrane fouling. This approach not only preserves membrane durability and function but also results in high performance of the electrolysis process.

SAMPLE

Sodium chloride brine, β(NaCl) = 300 g/L

EXPERIMENTAL

Add 10 mL oxidized sodium chloride brine sample and 2 mL of ultrapure water into the measuring vessel. The determination of iodide is carried out with the 884 Professional VA (Figure 1) using the parameters specified in Table 1. The concentration is determined by two additions of iodate standard addition solution.

Figure 1. 884 Professional VA manual for MME.

Table 1. Parameters

Parameter	Setting
Mode	HMDE
Start potential	-0.7 V
End potential	-1.3 V
Sweep rate	13 mV/s
Peak potential iodide	-1.05 V

ELECTRODES

- Multi-Mode Electrode pro

RESULTS

The chlor-alkali brine solution contains a high concentration of chloride ions that can interfere with the direct measurement of iodide. By converting iodide to iodate, these interferences are minimized. The determined iodate concentration is then recalculated as iodide concentration as indicated in Table 2.

The method is suitable for the determination of low concentrations of iodide in sodium chloride brine (β(NaCl) = 300 g/L) samples.

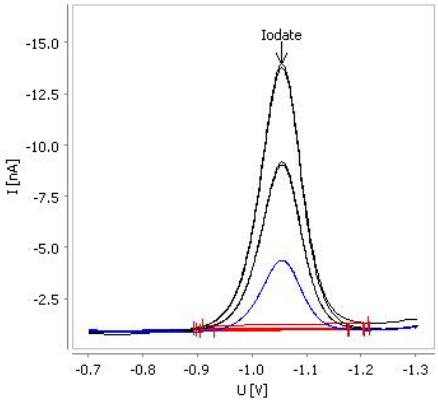

Figure 2. Determination of iodate in sodium chloride brine with stripping voltammetry.

Table 2. Result

Sample	Iodate (µg/L)
Sodium chloride brine	72.86

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Sample	Iodate (µg/L)
Sodium chloride brine	72.86
Sample	Iodide (µg/L)
Sodium chloride brine	52.63

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CONFIGURATION

884 Professional VA manual for Multi-Mode Electrode (MME)

884 Professional VA manual for Multi-Mode Electrode (MME) is the entry-level instrument for high-end trace analysis with voltammetry and polarography with the Multi-Mode Electrode pro or the scTRACE Gold or the Bismuth drop electrode. The proven Metrohm electrode methods in combination with a high-performance potentiostat/galvanostat and the extremely flexible viva software open up new perspectives for the determination of heavy metals. The potentiostat with a certified calibrator readjusts itself automatically before each measurement, thus guaranteeing maximum precision.

Determinations with rotating disc electrodes can also be performed with the instrument, e.g. determinations of organic additives in electroplating baths with "Cyclic Voltammetric Stripping" (CVS), "Cyclic Pulse Voltammetric Stripping" (CPVS), and chronopotentiometry (CP). The replaceable measuring head enables rapid changes between the various applications with different electrodes.

The viva software is required for control, data collection, and evaluation.

The 884 Professional VA manual for MME is supplied with extensive accessories and a measuring head for the Multi-Mode Electrode pro. Electrode set and viva license need to be ordered separately.