

VA Application Note No. V - 178

Title:	Determination of copper in sea water with the mercury film electrode (MFE)
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Summary:	The concentration of Cu in sea water is determined by anodic stripping voltammetry (ASV) in acetate buffer on a mercury film electrode (MFE). Gallium is added to overcome zinc interferences.
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Sample:	Sea water
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Sample preparation:	None
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Analysis of Cu		
Acetate buffer pH 4.6	$c(\text{CH}_3\text{COONH}_4) = 1 \text{ mol/L}$	
	$c(\text{CH}_3\text{COOH}) = 1 \text{ mol/L}$	
Mercury solution	$\beta(\text{Hg}^{2+}) = 1 \text{ g/L}$	
Gallium solution	$\beta(\text{Ga}^{3+}) = 10 \text{ mg/L}$	
Measuring solution	7.5 mL sea water + 2 mL acetate buffer pH 4.6 + 0.2 mL mercury solution + 0.05 mL gallium solution	
Working electrode (WE)	UT-RDE: drive shaft 6.1246.000 Ultra trace graphite electrode 6.1204.180	
Auxiliary electrode (AE)	GC electrode holder 6.1241.020 glassy carbon rod 6.1247.000	
Reference electrode (RE)	Reference system: Ag/AgCl/KCl (3 mol/L) Intermediate electrolyte: $c(\text{KCl}) = 3 \text{ mol/L}$	6.0728.020 6.1245.010
Parameters	Working electrode Stirrer speed Mode Purge time Conditioning Cycles Cleaning potential Cleaning time	RDE 2000 rpm DP 300 s Start potential: -0.5 V End potential: -0.1 V No. of cycles: 50 -0.15 V 20 s

Deposition potential	-0.8 V
Deposition time	75 s
Equilibration time	10 s
Pulse amplitude	0.05 V
Start potential	-0.6 V
End potential	-0.1 V
Voltage step	0.006 V
Voltage step time	0.1 s
Sweep rate	0.06 V/s
Peak potential Cu	-0.28 V

Results:	Cu
	15.6 µg/L

Determination of Cu

