

# VA Application Note No. V - 167

<b>Title:</b>	Cadmium in a phosphatation bath
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<b>Summary:</b>	The concentration of Cd in a Zn phosphatation bath is determined by polarography in HCl electrolyte.
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<b>Sample:</b>	Zn phosphatation bath
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<b>Sample preparation:</b>	None
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## Analysis of Cd

**HCl solution** w(HCl) = 30%

**Measuring solution** 10 mL H<sub>2</sub>O  
+ 1 mL HCl solution  
+ 1 mL phosphatation bath

**Working electrode (WE)** MME (Multi Mode Electrode) 6.1246.020

**Auxiliary electrode (AE)** Pt 6.0343.000

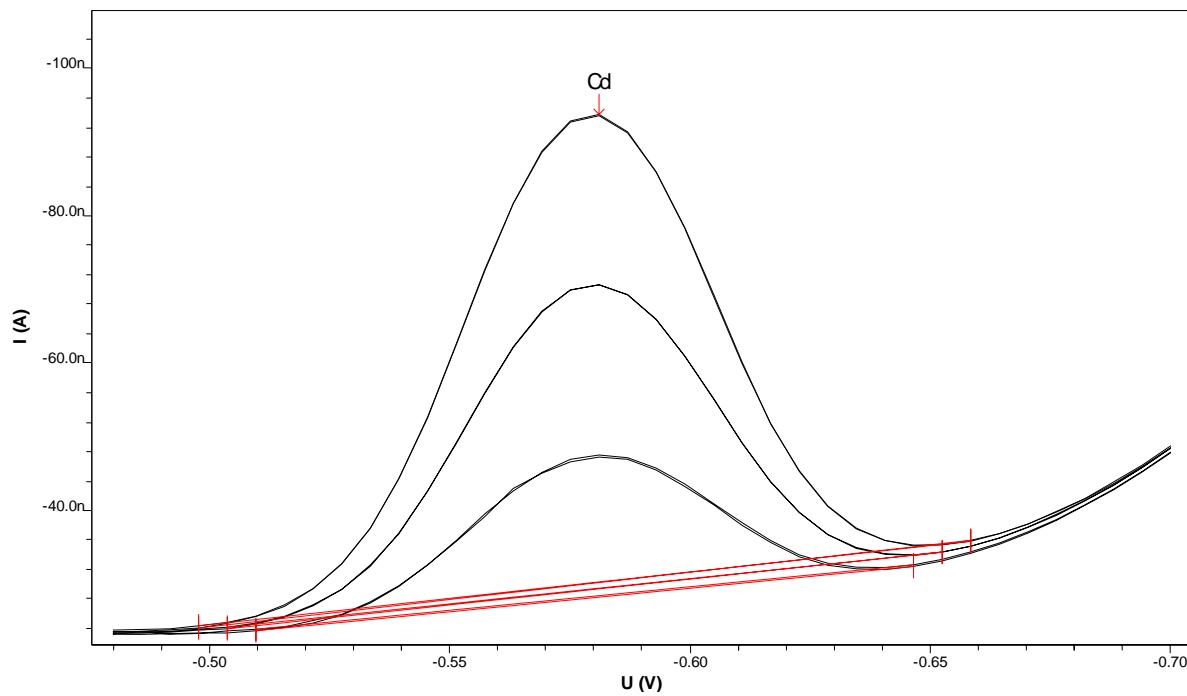
**Reference electrode (RE)** Reference system: Ag/AgCl/KCl (3 mol/L) 6.0728.020  
Intermediate electrolyte: c(KCl) = 3 mol/L 6.1245.010

**Parameters**

Working electrode	DME
Stirrer speed	2000 rpm
Mode	DP
Purge time	300 s
Equilibration time	5 s
Pulse amplitude	0.05 V
Start potential	-0.48 V
End potential	-0.7 V
Voltage step	0.006 V
Voltage step time	0.6 s
Sweep rate	0.01 V/s
Peak potential Cd	-0.58 V

<b>Results:</b>	Cd
	2.6 mg/L

## Determination of Cd



Cd  
c = 2.577 mg/L  
+/- 0.005 mg/L (0.20%)

