

# VA Application Note No. V - 160

<b>Title:</b>	Palladium in an activator bath
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<b>Summary:</b>	The concentration of Pd in activator bath is determined by polarography in ammonium chloride electrolyte.
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<b>Sample:</b>	Activator solution
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<b>Sample preparation:</b>	None
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## Analysis of Pd

**Supporting electrolyte**     $c(\text{NH}_4\text{Cl}) = 1 \text{ mol/L}$

$c(\text{NH}_3) = 1 \text{ mol/L}$

$c(\text{KCl}) = 1.5 \text{ mol/L}$

**Measuring solution**    15 mL  $\text{H}_2\text{O}$   
                              + 5 mL supporting electrolyte  
                              + 0.5 mL activator solution

**Working electrode (WE)**    MME (Multi Mode Electrode)    6.1246.020

**Auxiliary electrode (AE)**    Pt    6.0343.000

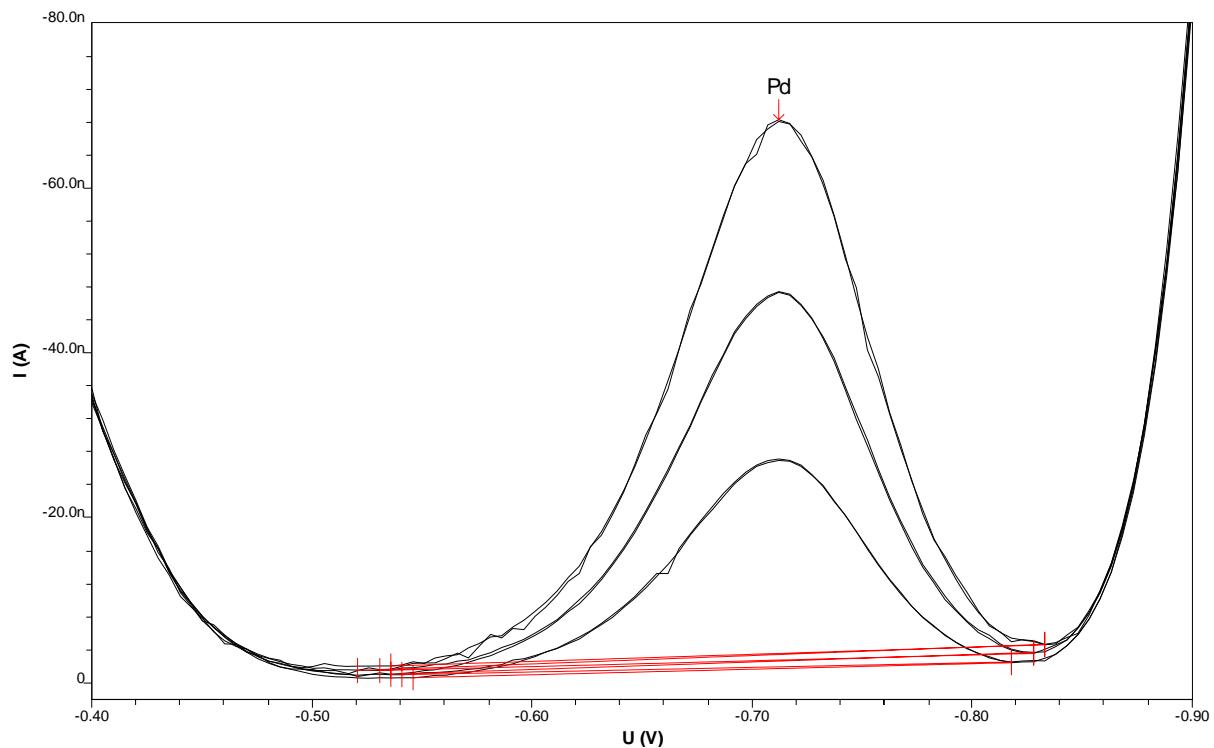
**Reference electrode (RE)**    Reference system: Ag/AgCl/KCl (3 mol/L)    6.0728.020  
                                    Intermediate electrolyte:  $c(\text{KCl}) = 3 \text{ mol/L}$     6.1245.010

**Parameters**

Working electrode	DME
Stirrer speed	2000 rpm
Mode	DP
Purge time	300 s
Equilibration time	5 s
Pulse amplitude	0.05 V
Start potential	-0.4 V
End potential	-0.9 V
Voltage step	0.005 V
Voltage step time	0.6 s
Sweep rate	0.0083 V/s
Peak potential Pd	-0.71 V

<b>Results:</b>	Pd
	64.1 mg/L

## Determination of



Pd  
c = 64.067 mg/L  
+/- 0.287 mg/L (0.45%)

