

VA Application Note No. V - 151

Title:	Antimony(III) and total antimony in electroless nickel bath
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Summary:	The concentration of Sb(III) and Sb(total) in an electroless nickel bath is determined by anodic stripping voltammetry (ASV). In c(HCl) = 0.6 mol/L only Sb(III) shows a signal. In w(HCl) = 10% the Sb(total) content is determined.
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Sample:	Electroless Ni bath
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Sample preparation:	None
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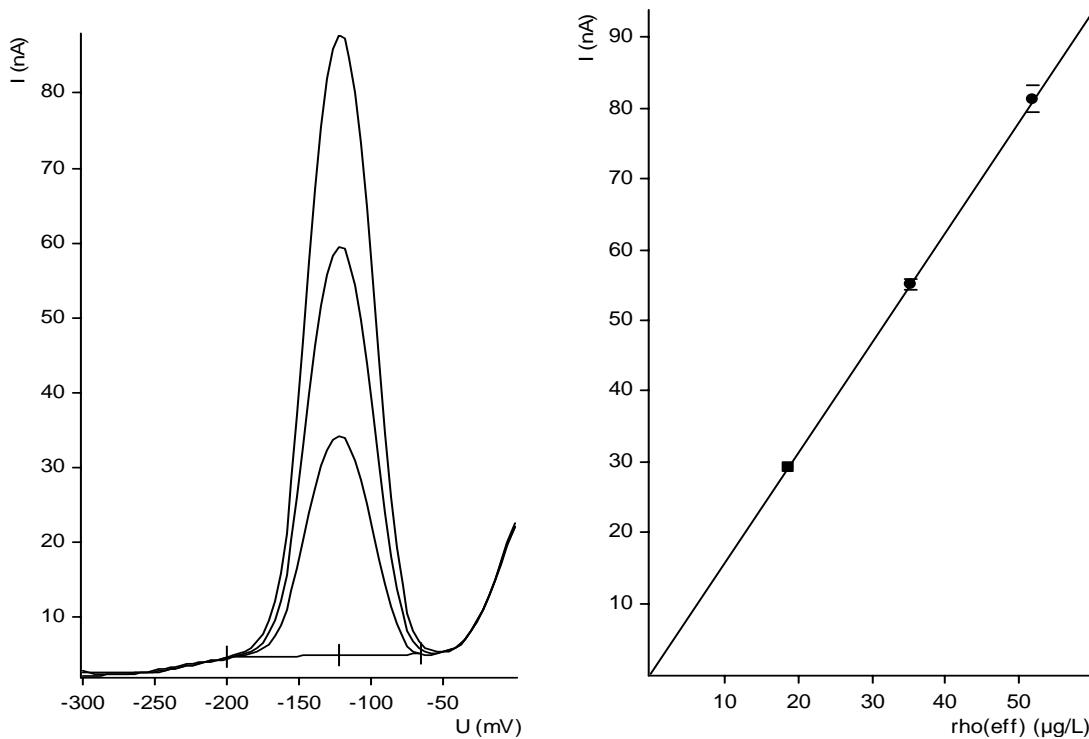
Analysis of Sb(III)		
Electrolyte for Sb(III)	c(HCl) = 0.6 mol/L	
Measuring solution	15 mL electrolyte for Sb(III) + 50 µL Ni plating bath	
Working electrode (WE)	MME (Multi Mode Electrode)	6.1246.020
Auxiliary electrode (AE)	Pt	6.0343.000
Reference electrode (RE)	Reference system: Ag/AgCl/KCl (3 mol/L) Intermediate electrolyte: c(KCl) = 3 mol/L	6.0728.020 6.1245.010
Parameters	Working electrode	HMDE
	Stirrer speed	2000 rpm
	Mode	DP
	Purge time	300 s
	Deposition potential	-0.45 V
	Deposition time	30 s
	Equilibration time	10 s
	Pulse amplitude	0.02 V
	Start potential	-0.3 V
	End potential	0 V
	Voltage step	0.004 V
	Voltage step time	0.3 s
Sweep rate		0.013 V/s
Peak potential Sb(III)		-0.12 V

Analysis of Sb(total)

Electrolyte for Sb(total) w(HCl) = 10%

Measuring solution	15 mL electrolyte for Sb(total) + 20 µL Ni plating bath	
Working electrode (WE)	MME (Multi Mode Electrode)	6.1246.020
Auxiliary electrode (AE)	Pt	6.0343.000
Reference electrode (RE)	Reference system: Ag/AgCl/KCl (3 mol/L) Outer system: c(KCl) = 3 mol/L	6.0728.020 6.1245.010
Parameters	Working electrode Stirrer speed Mode Purge time Deposition potential Deposition time Equilibration time Pulse amplitude Start potential End potential Voltage step Voltage step time Sweep rate Peak potential Sb(total)	HMDE 2000 rpm DP 300 s -0.45 V 30 s 10 s 0.02 V -0.3 V -0.1 V 0.004 V 0.3 s 0.013 V/s -0.19 V

Results:	Sb(III)	Sb(total)
	5.6 mg/L	47.3 mg/L

Determination of Sb(III)**Determination of Sb(total)**