

VA Application Note No. V - 115

Title: Antimony in PET (polyethylene terephthalate)

Summary: Sb is determined in polyethylene terephthalate (PET) after digestion in sulfuric acid and hydrogen peroxide. The application is carried out with anodic stripping voltammetry (ASV) in hydrochloric acid.

Sample: PET pellets

Sample preparation: 250 mg sample
2 mL w(H₂SO₄) = 96 % suprapur
are put into a flask of the digestion apparatus and heated to 250°C.
1 mL w(H₂O₂) = 30% suprapur is added through a dropping funnel. The mixture reacts intensively. Another 1 mL H₂O₂ is added. After the 2nd addition the mixture was heated to 400° C until SO₃ fumes are formed. After cooling down the solution was filled up to 100 mL.

Analysis of Sb

Electrolyte w(HCl) = 30 % suprapur

Measuring solution 5 mL water
+ 100 µL digestion solution
+ 5 mL w(HCl) 30%

Auxiliary electrode (AE) Pt

Reference electrode (RE) Ag/AgCl/KCl (3 mol/L)

Parameters

Working electrode	HMDE
Stirrer speed	2000 rpm
Mode	DP
Purge time	300 s
Deposition potential	-300 mV
Deposition time	30 s
Equilibration time	5 s
Pulse amplitude	50 mV

Start potential	-300 mV
End potential	-70 mV
Voltage step	6 mV
Voltage step time	0.1 s
Sweep rate	60 mV/s
Peak potential Sb	-200 mV

Results:	Sb
	158 $\mu\text{g/g}$

Determination of Sb

