

# Ti Application Note No. T- 28

<b>Title:</b>	<b>Hydrogen sulphide and mercaptans in petroleum products</b>
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<b>Summary:</b>	Simultaneous determination of hydrogen sulphide and mercaptans in petroleum products by potentiometric titration with silver nitrate using the Ag Titrode.
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<b>Sample:</b>	Petroleum products, e.g. diesel fuel, petrol, gasoline etc.
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<b>Sample Preparation:</b>	none
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<b>Instruments and Accessories:</b>	702, 716 or 736 Titrino or 726 Titroprocessor, 6.0430.100 Ag Titrode with Ag <sub>2</sub> S coating
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<b>Analysis:</b>	Pour 100 mL acetone, 20 mL isopropanol and 5 mL electrolyte (prepared by dissolving 20 g NH <sub>4</sub> NO <sub>3</sub> in 100 mL NH <sub>3</sub> 25%) into the titration vessel and deaerate the solution by passing nitrogen through it. Add 3.5 ... 11 g sample (exactly weighed) and titrate under nitrogen with c(AgNO <sub>3</sub> ) = 0.0005 mol/L.
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<b>Calculation:</b>	<p>If the sample contains both H<sub>2</sub>S and mercaptans, two equivalence points are obtained, the first of which corresponds to the H<sub>2</sub>S and the second to the mercaptans.</p> <p><math display="block">\text{ppm H}_2\text{S sulphur} = \text{EP1} * \text{C01} * \text{C02} / \text{C00}</math></p> <p><math display="block">\text{ppm mercaptan sulphur} = (\text{EP2} - \text{EP1}) * \text{C03} * \text{C02} / \text{C00}</math></p> <p>EP1 = titrant consumption in mL to reach the first EP EP2 = titrant consumption in mL to reach the second EP C00 = sample weight in g C01 = 0.008 (H<sub>2</sub>S sulphur equivalent in mg/mL; 1 mL c(AgNO<sub>3</sub>) = 0.0005 mol/L corresponds to 0.008 mg H<sub>2</sub>S sulphur) C02 = 1000 (conversion factor for ppm) C03 = 0.016 (mercaptan sulphur equivalent in mg/mL; 1 mL c(AgNO<sub>3</sub>) = 0.0005 mol/L corresponds to 0.008 mg mercaptan sulphur)</p>
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**Remarks:** Before adding the sample the solvent mixture has to be absolutely free of oxygen!

**Results:**

AVG(3) = 2.6 +/- 0.1 ppm H<sub>2</sub>S sulphur

AVG(3) = 32.28 +/- 0.1 ppm mercaptan sulphur