Thermo. Titr. Application Note No. H-052

Title:	Determination of nickel in electroless plating
	solutions

Scope:	Thermometric	titration	of	nickel	in	electroless	plating
	solution with disodium dimethylglyoximate.					_	

Principle:	Titration of Ni(II) with standard sodium dimethylglyoximate (DMG) solution in buffered ammonia solution to an exothermic endpoint. Two moles of DMG react with one mole of Ni(II). Acidic Ni solutions should be complexed
	with citrate prior to basification.

Reagents:	Titrant: 0.5 mol/L disodium dimethylglyoximate. Dissolve 153.6 g disodium dimethylglyoximate (99% pure, FW = 304.21) in deionized water and make to 1000mL in a volumetric flask.
	NH_3/NH_4CI buffer: Dissolve 17.5 g A.R. NH_4CI in 172 mL A.R. conc. NH_3 soln. Make to 250 mL with deionized water.

Method:	Basic Experimental Parameters:	Basic Experimental Parameters:		
	Titrant delivery rate (mL/min.)	2		
	No. of exothermic endpoints	1		
	Data smoothing factor	55		
	Stirrer speed	10-12*		
	Procedure: Pipette a aliquots of M approximately 0.3 – 1 mmole Ni Add 2mL NH ₃ /NH ₄ Cl buffer solu- water to give a starting volume ran 0.3 mmole Ni, and 50mL for 1 mm of Ni, employ a stirring speed amounts a speed of 12 (the sus viscous with higher amounts of stirring speeds* and larger than u Titrate to a single exothermic e disodium dimethylglyoximate.	Ni solution ranging from (II) into titration vessel. ution, and sufficient DI nging between 35mL for hole Ni. For low amounts of 10, and for higher spension becomes quite Ni, hence the higher usual volumes of water).		
	Equipment stained by the precipitate may be cleaned by so strong mineral acid; eg, 1 mol/L H0	Ni dimethylglyoximate baking in a solution of a Cl.		

The exposed thermistor may be cleaned periodically with a soft toothbrush, although it is also rapidly cleaned by soaking in 1 mol/L HCI.

Results:Titration of aliquots of electroless plating solution
containing nickel and sodium hypophosphiteNi = 29.51±0.03g/L (%RSD = 0.11, n=6)

