

Application Bulletin 266/2 e

Determination of titanium by adsorptive stripping voltammetry

Summary

This Application Bulletin describes the determination of titanium by adsorptive stripping voltammetry (AdSV) using mandelic acid as complexing agent. The method is suitable for the analysis of ground, drinking, sea, surface and cooling waters, in which the concentration of titanium is of importance. The methods can, of course, also be used for the trace analysis in other matrices.

The limit of detection is approx. $0.5 \mu g/L$.

Instruments

VA instrument	
capable of operating a Multi-Mode	
Electrode and supporting differential	
pulse (DP) measuring mode	
909 UV Digester	2.909.0014

Electrodes

WE	Multi-Mode Electrode pro Mercury drop capillary	6.1246.120 6.1226.030 or 6.1226.050
RE	Ag/AgCl reference electrode Ag/AgCl/KCl (3 mol/L)	6.0728.x20 6.1245.010
	Electrolyte vessel Filled with c(KCI) = 3 mol/L	0.1245.010
AE	Pt rod electrode	6.0343.x00

Reagents

All of the used reagents must be of purest quality possible (for analysis or for trace analysis*).

- DL-mandelic acid, for analysis, CAS 611-72-3
- Ammonia solution, w(NH₃) = 25 %, for trace analysis*, CAS 1336-21-6
- Ti standard stock solution, β(Ti) = 1 g/L (commercially available)
- Ultrapure water, resistivity >18 MΩ·cm (25 °C), type I grade (ASTM D1193)

* e.g., Merck suprapur®, Honeywell Fluka TraceSelect® or equivalent

Solutions

Mandelic acid solution	c(mandelic acid) = 0.4 mol/L 6.1 g mandelic acid are dissolved in 100 mL ultrapure water. The solution is stable for 1 week.
Diluted ammonia solution	w(NH ₃) = 10%

Standard solutions

Ti standard solution	$\beta(Ti) = 1 \text{ mg/L}$
	The standard stock solution is
	diluted with c(HCI) = 0.1 mol/L.
	The standard has to be prepared
	fresh daily.

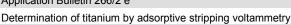
Sample preparation

 Ground water, surface waters, mineral waters and drinking waters can usually be analyzed without pretreatment.

Organic matter often interferes with voltammetric determinations and therefore sample solutions usually have to be digested.

- Low polluted waste waters can be digested with the 909 UV Digester.
 - ο Add 50 100 μ L w(H₂O₂) = 30% and 10 μ L w(HCI) = 30% to 10 mL acidified sample (pH = 2) and irradiate for 90 min at 90 °C. After cooling to room temperature, the digested sample can be transferred directly to the polarographic vessel.
- Samples with organic matter (foods, pharmaceuticals etc.) must be digested.
 - o High-pressure asher
 - o Microwave digestion

Both techniques oxidize the samples in a closed digestion vessel by means of a mixture of concentrated mineral acids.





Analysis

Measuring solution

10 mL (diluted) sample

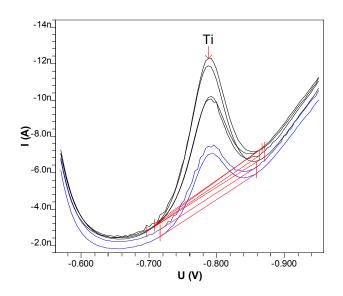
1 mL mandelic acid solution

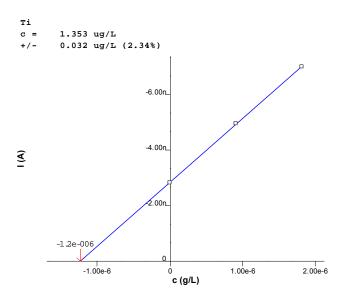
Adjust the pH value to 3.0 with $w(NH_3) = 10\%$.

Parameters

Voltammetric	
Electrode operating mode	HMDE
Measuring mode	DP – Differential pulse
Stirring rate	2000 min ⁻¹
Potentiostatic pretreatment	
Potential 1	-0.57 V
Waiting time 1	30 s
Equilibration time	10 s
Sweep	
Start potential	-0.57 V
End potential	-0.95 V
Potential step	0.004 V
Potential step time	0.2 s
Sweep rate	0.02 V/s
Pulse amplitude	0.05 V
Substance	
Name	Ti
Characteristic potential	-0.79 V

Example





Result

Sample	Sea water
Sample size	10.0 mL
β(Τί)	1.4 μg/L

References

- Van den Berg, C. M. G., Potentials and potentialities of cathodic stripping voltammerty of trace elements in natural waters, Anal. Chim. Acta (1991) 250: 265-276
- Yokoi, K., Van den Berg, C. M. G., Determination of titanium in sea water using catalytic cathodic stripping voltammetry, Anal. Chim. Acta (1991) 245: 2/ 167-176
- Li, H., Van den Berg, C. M. G., Determination of Titanium in sea water using adsorptive cathodic stripping voltammetry, Anal. Chim. Acta (1989) 221/2 269-277
- Metrohm Application Bulletin 113



Appendix

Report for the example determination titanium in seawater

```
======== METROHM 757 VA COMPUTRACE (5.757.0020) =============
            : 11021144_Sea Water Ti.dth
: Sea Water Ti
 Sample ID
                                  Date : 1999-11-02
 Creator
                                                               Time:
                                                                       11:44:42
                                  Date : 2001-07-05
Modified by : zu
                                                               Time:
                                                                       08:43:10
     : zu
                                  Date : 2001-07-05
                                                               Time: 08:43:11
User
 Cell volume: 11.000 mL
Sample amount: 10.000 mL
Method : Ti Determination.mth
Title : Ti in Sea Water
Remark1 : 10 mL Seawater + 1mL Mandelic Acid 0.4 mol/L
Remark2 : NH3 10% --->pH 3
  Substance : Ti
                                                                Comments
  Mass conc.:
                 1.230 ug/L
               0.031
13.534
                                  ( 2.55%)
 MC.dev. :
                         ug/L
 Mass
                        ng
               10.000 ng
 Add.mass :
              VR
                     V
                                   I.mean Std.Dev. I.delta
                   -0.788 -2.785
                                   -2.824 0.055
             1 - 1
                   -0.788 -2.863
-0.788 -4.916
             1 - 2
             2-1
                                   -4.954 0.053
                                                     -2.130
                   -0.788 -4.992
             2-2
                    -0.788 -7.024
                                    -7.003 0.050
                    -0.788 -6.982
 Substance
              Calibr.
                          Y.reg/offset
                                          Slope
                                                        Std.Dev.
              std.add.
                          -2.837e-009
                                          -2.305e-003
                                                       5.214e-011
 Final results
                                         +/- Res. dev. %
Ti:
                        1.353 µg/L
                                         0.035
                                                      2.554
 Titanium
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Method print for the determination of titanium

Method paramet	Method parameters			
Title	: AB266 Det of Ti.mth : Determination of Titanium. AB266 method 1 : 10 mL sample + 1 mL mandelic acid> pH 3 :			
Calibration Technique Addition				
Sample ID : sample Sample amount (mL): 10.000 Cell volume (mL): 11.000				
Voltammetric parameters				
Mode		: DP - Differential Pulse		
Highest current Lowest current		: 10 mA : 100 nA		
Electrode Drop size (19 Stirrer speed (: HMDE : 4 : 2000		
Initial electr.	conditioning	: No		
No. of addition		: 2 : 2		



Measure b Addition	olank purge time (s)	: No : 10		
Initial p	purge time (s)	: 3	00	
		: -1.2 : -0.1	00	
Cleaning Cleaning Deposition		: No : -0.1 : 0.0 : -0.5 : 30.0	00 00 70	
Start pot End poter Voltage s Voltage s Sweep rat	step time (s) ce (V/s) plitude (V)	: 5.00 : -0.55 : -0.9 : 0.00 : 0.00 : 0.00 : 0.00	70 50 04 00 20 50	
Cell off	after measurement	: Ye	es	
Peak eval				
Regressic Peak eval Minimum p Minimum p Reverse p Smooth fa Eliminate	on technique Luation peak width (V.steps) peak height (A) peaks actor	: Linear : Height : 5 : 1.000e : No : 4 : Yes	Regression	
Substance				
Ti	: -0.7	0 V +/- 0		
Standard Addition	solution : 1 volume (mL) : 0.01	.000 mg/L		
Titanium		result (Ti * (11 / 10)) = * (1e+006 / 1)	+ 0 - 0
Baseline				
Substance	Addition automatic	start (V) e	nd (V) type	scope
Ti 	Sample yes Addition 1 yes Addition 2 yes	 	linear linear linear	wholePeak wholePeak wholePeak