

## Application Bulletin 199/4 e

# Determination of sulfide and sulfite by polarography

### Summary

Sulfide and sulfite can be determined polarographically without any problems. For sulfide, polarography is performed in an alkaline, for sulfite in a slightly acidic primary solution. The method is suitable for the analysis of pharmaceuticals (infusion solutions), wastewater, flue gas water, photographic solutions, etc.

### Instruments

VA instrument  
capable of operating a Multi-Mode Electrode and  
supporting differential pulse (DP) measuring mode

### Electrodes

WE	Multi-Mode Electrode pro	6.1246.120
	Mercury drop capillary	6.1226.030
RE	Ag/AgCl reference electrode	6.0728.x20
	Ag/AgCl/KCl (3 mol/L)	
	Electrolyte vessel	6.1245.010
	Filled with c(KCl) = 3 mol/L	
AE	Pt rod electrode	6.0343.x00

## Method 1: Sulfide determination

### Reagents

All of the used reagents must be of analysis quality (for analysis).

- Sodium hydroxide, w(NaOH) = 30 %, for analysis, CAS 1310-73-2
- Sodium sulfide, Na<sub>2</sub>S, for analysis, CAS 27610-45-3
- Ultrapure water, resistivity >18 MΩ·cm (25 °C), type I grade (ASTM D1193)

### Solutions

Diluted NaOH solution      c(NaOH) = 0.1 mol/L  
5 mL sodium hydroxide is made  
up to 500 mL with ultrapure water.

### Standard solutions

Sulfide standard solution      β(sulfide) = 1 g/L  
The standard solution is prepared  
from Na<sub>2</sub>S with oxygen free  
sodium hydroxide solution  
c(NaOH) = 0.1 mol/L.

### Analysis

#### Measuring solution

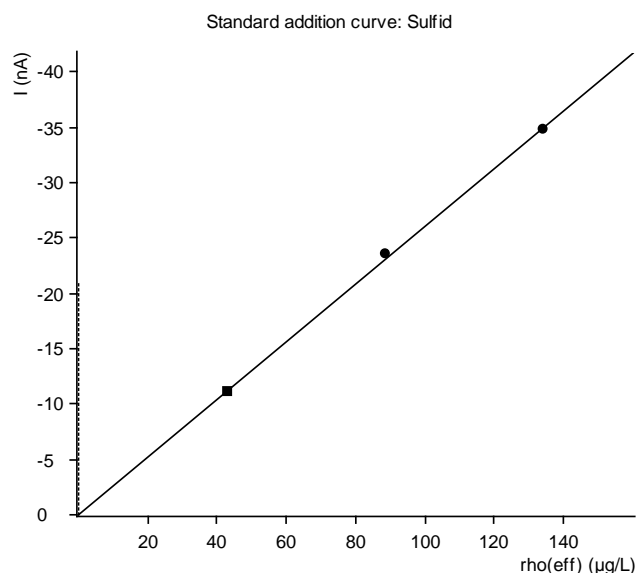
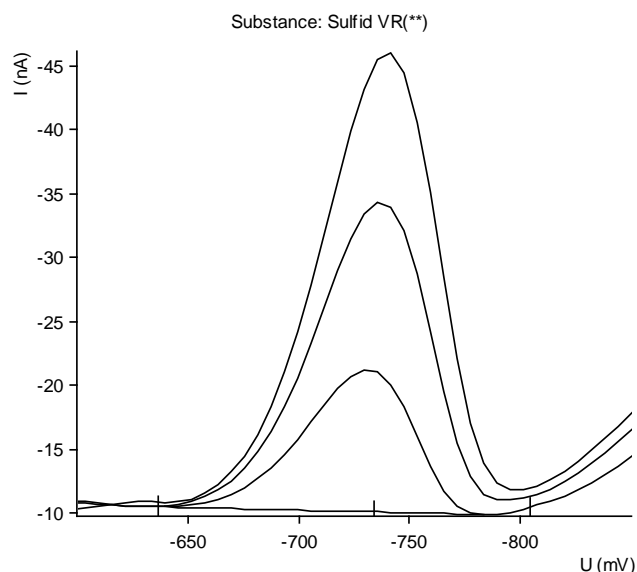
10 mL diluted sodium hydroxide solution  
(purge for 5 min with nitrogen)  
10 mL (diluted) sample  
(mix while stirring without nitrogen)

The concentration is determined by standard addition.

### Parameters

Voltammetric	
Electrode operating mode	SMDE
Measuring mode	DP – Differential pulse
Stirring rate	2000 min <sup>-1</sup>
Equilibration time	10 s
Sweep	
Start potential	-0.5 V
End potential	-0.9 V
Potential step	0.006 V
Potential step time	0.6 s
Sweep rate	0.01 V/s
Pulse amplitude	0.05 V
Substance	
Name	Sulfide
Characteristic potential	-0.75 V

## Example



## Result

Sample size	1.0 mL
$\beta$ (sulfide)	475.2 µg/L

## Comments

- When the 797 VA Computrace is used the purging has to be done manually prior to the start of the method.
- After the addition of a sample solution or standard solutions, purging with nitrogen may no longer be performed, otherwise loss of sulfide could occur.
- A linearity test was performed between 0.02 and 2 mg/L: Between 0.02 and 0.25 mg/L at the SMDE and between 0.25 mg/L and 2 mg/L at the DME (bigger Hg-

drop). The sulfide determination is linear up to 1.6 mg/L.

- The determination limit for sulfide lies by 20 µg/L.

## Method 2: Sulfite determination

### Reagents

All of the used reagents must be of analysis quality (for analysis).

- Sodium hydroxide, for analysis,  $w(\text{NaOH}) = 30\%$ , CAS 1310-73-2
- Acetic acid, for analysis,  $w(\text{CH}_3\text{COOH}) = 100\%$ , CAS 64-19-7
- Sodium sulfite,  $\text{Na}_2\text{SO}_3$ , for analysis, CAS 7757-83-7
- Ultrapure water, resistivity  $>18 \text{ M}\Omega \cdot \text{cm}$  (25 °C), type I grade (ASTM D1193)

### Solutions

Acetate buffer (pH 4.6)	$c(\text{NaOH}) = 0.2 \text{ mol/L}$ , $c(\text{CH}_3\text{COOH}) = 0.4 \text{ mol/L}$ 10 mL sodium hydroxide and 11.1 mL acetic acid are made up to 500 mL with ultrapure water.
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### Standard solutions

Sulfite standard solution	$\beta(\text{sulfite}) = 1 \text{ g/L}$ The standard solution is prepared from $\text{Na}_2\text{SO}_3$ with oxygen free ultrapure water.
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### Analysis

#### Measurement solution

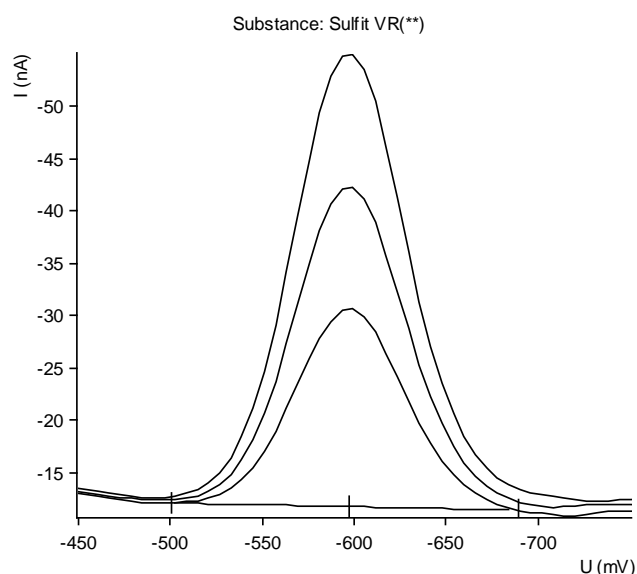
- 10 mL acetate buffer
- (deaerate 5 minutes with nitrogen)
- 10 mL (diluted) sample
- (without degassing with nitrogen, stir for 10 seconds)

The concentration is determined by standard addition.

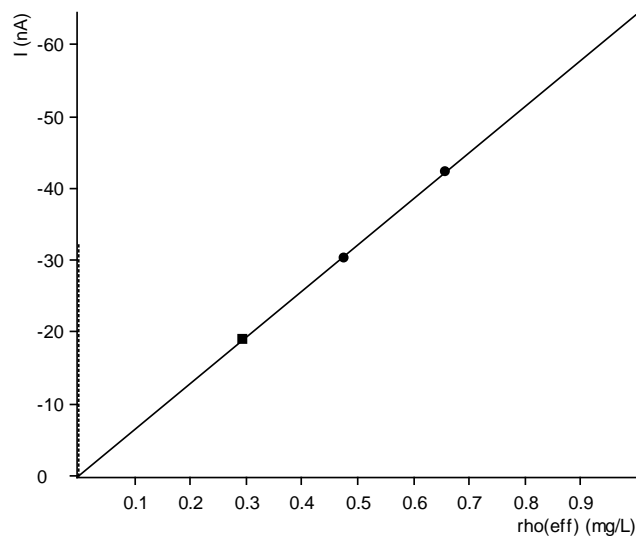
## Parameters

Voltammetric	
Electrode operating mode	DME
Measuring mode	DP – Differential pulse
Stirring rate	2000 min <sup>-1</sup>
Equilibration time	10 s
Sweep	
Start potential	-0.4 V
End potential	-0.85 V
Potential step	0.006 V
Potential step time	0.4 s
Sweep rate	0.015V/s
Pulse amplitude	0.05 V
Substance	
Name	Sulfite
Characteristic potential	-0.6 V

## Example



Standard addition curve: Sulfite



## Result

Sample size	1.0 mL
$\beta$ (sulfite)	3.2 mg/L

## Comments

- With the 797 VA Computrace the purging has to be done manually before the start of the method.
- After addition of a sample solution or standard solution, purging with nitrogen may no longer be performed, otherwise sulphur dioxide can escape. (mix only with stirring).
- In the presence of sulfide, a peak appears at -0.45 V. This cannot be used for quantitative analyses. (Acidic medium - high volatility of hydrogen sulfide).
- Sulfite can also be determined in 1 mmol/L hydrochloric acid as supporting electrolyte.
- Should sulfide and sulfite be determined together in the same sample, the sulfide must first be determined in alkaline solution, and after addition of 250  $\mu$ L 50% acetic acid / 10 mL acetate buffer, sulfite can be analysed.
- In the presence of thiosulfate, two overlapping peaks can be observed between -0.14 V and -0.28 V. With thiosulfate contents up to ca. 100  $\mu$ g/initial mass, only one peak appears at -0.28 V. This can be used for quantitative analyses.
- The determination of sulfite should be performed immediately upon taking the sample, because sulfite solutions are not stable.

## References

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- Cheshchevoi, V.N. / Polushkin, V.A. / Slovetskii, V.L., Analysis of the mixture of sulphur containing ions in water solutions by polarography, *Zavodsk. Lab.* 51 (1985), 16-17
- Donahue, J.J. / Oliveri- Vigh, S., Simultaneous polarographic determination of bisulfite and iproniazid in gel formulations, *J. Pharm. Sci.* 62, (1973), 1990-1992
- Garber, R.W. / Wilson, C.E., Determination of atmospheric sulfur dioxide by differetial pulse polarography, *Anal. Chem.* 44, (1972), 1357-1360
- Holak, W. / Patel, B., Differential pulse polarographic determination of sulfites in food: Collaborative study, *J. Assoc. Off. Anal. Chem.* 70, (1987), 572-578
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## Appendix

### Report for the example determination of sulfide according to method 1

```
===== METROHM 746 VA TRACE ANALYZER (5.746.0101) =====
Determin. : AD_DME_S User: Date: 1993-05-21
Modified : 1993-05-21 09:44:37 Run : 0 Time: 09:33:08
Sample table: -
```

Pos.	Ident.1/S1 Standard	Ident.2/S2	Ident.3/S3	Method.call	Sample size/S0 1 mL
-----					
Method	A199S_A2				
Title	Bestimmung von Sulfid an der DME mittels Std.Add.				
Remark1	Bestimmung von Sulfid mittels Standardaddition				
Remark2					

Substance	Mass conc.	MC.dev.	Cal.dev.	Mass	Add.mass	V0.sample	Comments
Sulfid	475.2 ug/L	54.5 ug/L (11.5%)	-	475.2 ng	500 ng	1 mL	
-----							
VR	U/mV	I/nA	I.mean	Std.dev.	I.delta	Comments	
00	-734	-11.07	-11.07				
10	-740	-23.38	-23.38		-12.31		
20	-743	-34.44	-34.44		-11.06		

Substance	Techn.	Y.reg/offset	Slope	Nonlin.	Mean deviat.
Sulfid	std.add.	-1.126e-08	-2.606e-04		4.715e-10

Final results	+/-	Res.dev.	%	Comments
Sulfid = 475.24 ug/L	54.5	11.5		

### Method print for the determination of sulfide according to method 1

```
===== METROHM 746 VA TRACE ANALYZER (5.746.0100) =====
Method: AB199_1 .mth OPERATION SEQUENCE
Title : Determination of Sulfide with SMDE. AB199 Part 1
```

	Instructions	t/s	Main parameters	Auxiliary parameters
1	DOS/M		V.added 10.000 mL	
2	REM		10 mL 0.1 mol/L NaOH	
3	PURGE			
4	STIR	300.0	Rot.speed 2000 /min	
5	OPURGE			
6	SMPL>M		V.fraction mL	V.total L
7	(ADD			
8	NOP	10.0		
9	SEGMENT		Segm.name pol	
10	ADD>M		Soln.name S-Std	V.add 0.050 mL
11	ADD)2			
12	END			

```
Method: AB199_1 SEGMENT
pol
```

	Instructions	t/s	Main parameters	Auxiliary parameters
1	OSTIR	3.0		
2	SMDE		Drop size 4	
3	DPMODE		U.ampl -50 mV	t.meas 30.0 ms
			t.step 0.60 s	t.pulse 40.0 ms
4	SWEEP	42.0	U.start -500 mV	U.step 6 mV
			U.end -900 mV	Sweep rate 10 mV/s
5	STIR		Rot.speed 2000 /min	
6	OMEAS		U.standby mV	
7	END			

## Report for the example determination of sulfite according to method 2

```
===== METROHM 746 VA TRACE ANALYZER (5.746.0101) =====
Determ.      : Adldmeso      User: zu      Date: 1993-05-23
Modified     : 1993-05-23 14:00:50 Run : 0    Time: 13:59:08
Sample table: -
```

Pos.	Ident.1/S1 Standard	Ident.2/S2	Ident.3/S3	Method.call	Sample size/S0 1 mL
Method : A199SOA1					
Title : Bestimmung von Sulfit an der DME mittels Std.Add.					
Remark1 : Bestimmung von Sulfit mittels Standardaddition					
Remark2 : Auswertung linear					

```
Substance : Sulfit
Mass conc.: 3.227 mg/L      Mass      : 3.227 ug
MC.dev.   : 0.123 mg/L (3.82%) Add.mass : 2 ug
Cal.dev.  : -              V0.sample: 1 mL
```

VR	U/mV	I/nA	I.mean	Std.dev.	I.delta	Comments
00	-597	-18.92	-18.92			
10	-597	-30.29	-30.29		-11.37	
20	-597	-42.11	-42.11		-11.83	

Substance	Techn.	Y.reg/offset	Slope	Nonlin.	Mean deviat.
Sulfit	std.add.	-1.883e-08	-6.421e-05		2.027e-10

```
Final results      +/- Res.dev.  %      Comments
-----
Sulfit = 3.2266 mg/L      0.123  3.82
```

## Method print for the determination of sulfite according to method 2

```
===== METROHM 746 VA TRACE ANALYZER (5.746.0100) =====
Method: AB199_2 .mth      OPERATION SEQUENCE
Title : Determination of Sulfite with DME. AB199 Part 2
```

	Instructions	t/s	Main parameters	Auxiliary parameters
1	DOS/M		V.added 10.000 mL	
2	REM		10 mL buffer (0.2 mol/L NaOH, 0.4 mol/L acetic acid)	
3	PURGE			
4	STIR	300.0	Rot.speed 2000 /min	
5	OPURGE			
6	SMPL>M		V.fraction mL	V.total L
7	(ADD			
8	NOP	10.0		
9	SEGMENT		Segm.name pol	
10	ADD>M		Soln.name SO3-Std	V.add 0.020 mL
11	ADD)2			
12	END			

```
Method: AB199_2      SEGMENT
                      pol
```

	Instructions	t/s	Main parameters	Auxiliary parameters
1	OSTIR			
2	DME			
3	DPMODE		U.ampl -50 mV	t.meas 20.0 ms
			t.step 0.40 s	t.pulse 40.0 ms
4	SWEEP	31.2	U.start -400 mV	U.step 6 mV
			U.end -850 mV	Sweep rate 15 mV/s
5	STIR		Rot.speed 2000 /min	
6	OMEAS		U.standby mV	
7	END			