

Determination of trace amounts of molybdenum (or tungsten) in water by polarography

Summary

Molybdenum is an essential trace element for plant growth. Since it occurs in natural waters only in trace amount, a very sensitive method of determination is needed. Using the following polarographic method, it is possible to determine $5 \cdot 10^{-10}$ mol/L resp. 50 ng/L.

The principle of the method is based on the reaction between the molybdate ion MoO_4^{2-} and the complexing agent 8-hydroxy-7-iodo-quinoline-5-sulfonic acid (H_2L) to form a $\text{MoO}_2\text{L}_2^{2-}$ complex, which is adsorbed on the mercury electrode. The adsorbed Mo(VI) is reduced electrochemically to the Mo(V) complex. The hydrogen ions present in the solution oxidize Mo(V) again spontaneously to form the Mo(VI) complex, which is thus newly available for electrochemical reduction. This catalytic reaction is the reason for the high sensitivity of the method.

Tungsten W(VI) exhibits practically the same electrochemical behavior as molybdenum, but is not described in detail in this Application Bulletin.

Instruments

VA instrument
capable of operating a Multi-Mode Electrode and
supporting differential pulse (DP) measuring mode

Electrodes

WE	Multi-Mode Electrode pro	6.1246.120
	Mercury drop capillary	6.1226.030
	or	6.1226.050
RE	Ag/AgCl reference electrode	6.0728.x20
	Ag/AgCl/KCl (3 mol/L)	
	Electrolyte vessel	6.1245.010
	Filled with $c(\text{KCl}) = 3$ mol/L	
AE	Pt rod electrode	6.0343.x00

Reagents

All of the used reagents must be of purest quality possible (for analysis or for trace analysis*).

- 8-hydroxy-7-iodo-quinoline-5-sulfonic acid, CAS 547-91-1
- Sulfuric acid, for trace analysis*, $w(\text{H}_2\text{SO}_4) = 96\%$, CAS 7664-93-9
- Potassium chloride, for trace analysis*, CAS 7447-40-7
- Mo standard stock solution: $\beta(\text{Mo}^{6+}) = 1$ g/L (commercially available)
- Ultrapure water, resistivity $>18 \text{ M}\Omega \cdot \text{cm}$ (25 °C), type I grade (ASTM D1193)

* e.g., Merck suprapur®, Honeywell Fluka TraceSelect® or equivalent

Solutions

Diluted sulfuric acid	$c(\text{H}_2\text{SO}_4) = 0.1$ mol/L 5.5 mL sulfuric acid are diluted to 1 L with ultrapure water.
Reagent solution	$c(8\text{-hydroxy-7-iodo-quinoline-5-sulfonic acid}) = 2 \cdot 10^{-4}$ mol/L $c(\text{KCl}) = 0.7$ mol/L $c(\text{H}_2\text{SO}_4) = 0.1$ mol/L Dissolve 35 mg 8-hydroxy-7-iodo-quinoline-5-sulfonic acid in 450 mL diluted sulfuric acid. Add 26.1 g potassium chloride and fill up to 500 mL with diluted sulfuric acid.

Standard solution

Molybdenum standard solution	$\beta(\text{Mo}^{6+}) = 100$ µg/L Prepare more diluted standard solutions $\beta(\text{Mo}^{6+}) = 10 \dots 200$ µg/L through dilution of the 1 g/L molybdenum standard stock solution with sulfuric acid 0.1 mol/L.
------------------------------	--

Sample preparation

- Ground water, sea water, mineral waters and drinking waters can usually be analyzed without pretreatment.

Organic matter often interferes with voltammetric determinations and therefore sample solutions usually have to be digested.

- Low polluted waste waters can be digested with the 909 UV Digester.
 - Add 50 μL hydrogen peroxide solution $w(\text{H}_2\text{O}_2) = 30\%$ and 10 μL hydrochloric acid $w(\text{HCl}) = 30\%$ to 10 mL acidified sample ($\text{pH} = 2$) and irradiate for 90 minutes at 90°C .
- Samples with organic matter (foods, pharmaceuticals etc.) must be digested.
 - High-pressure asher
 - Microwave digestion
 - Both techniques oxidize the sample in a closed digestion vessel by means of a mixture of concentrated mineral acids.
 - Open wet digestion with H_2SO_4 and H_2O_2 according to Application Bulletin 113.

Analysis

Measuring solution

10 mL (diluted) sample

2 mL reagent solution

The concentration is determined by standard addition.

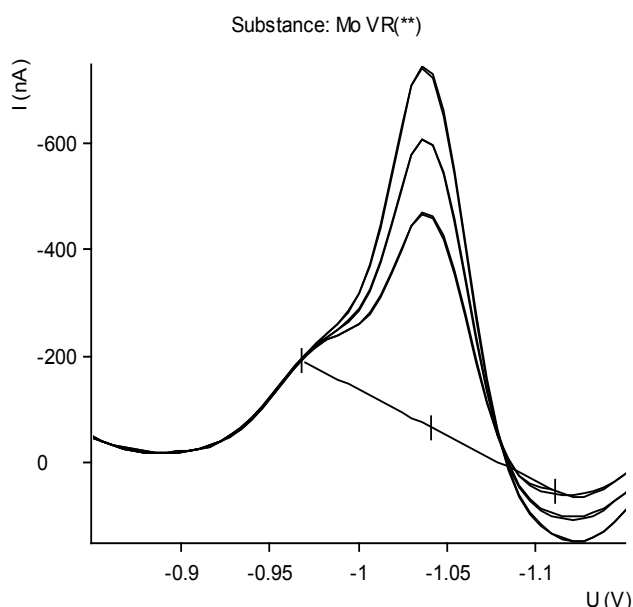
For an accurate determination of the concentration, the chemical blank value must be taken into consideration. To do this, determine the molybdenum concentration of 10 mL high purity water and 2 mL reagent solution under the same conditions as is given for the sample.

Parameters

Voltammetric	
Electrode operating mode	SMDE
Measuring mode	DP – Differential pulse
Stirring rate	2000 min^{-1}
Equilibration time	5 s
Sweep	
Start potential	-0.68 V
End potential	-1.18 V
Potential step	0.006 V
Potential step time	0.6 s
Sweep rate	0.01 V/s

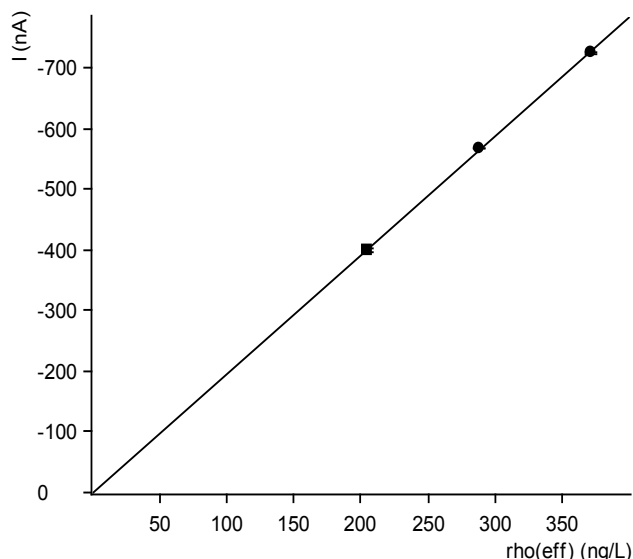
Pulse amplitude	0.05 V
Substance	
Name	Mo
Characteristic potential	-1.0 V

Example



The shoulder at - 980 mV is caused by approx. 250 $\mu\text{g/L}$ Zn

Standard addition curve: Mo



Results

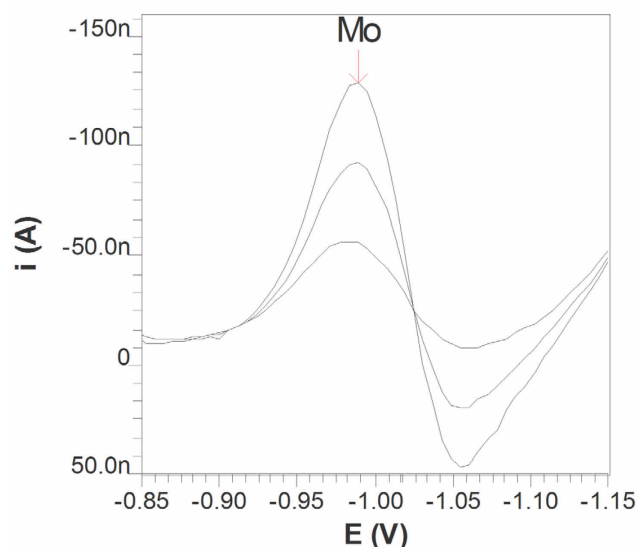
Sample	Drinking water
Sample size	10.0 mL
$\beta(\text{Mo})$	245 ng/L

Comments

Reduction of the reagents blank

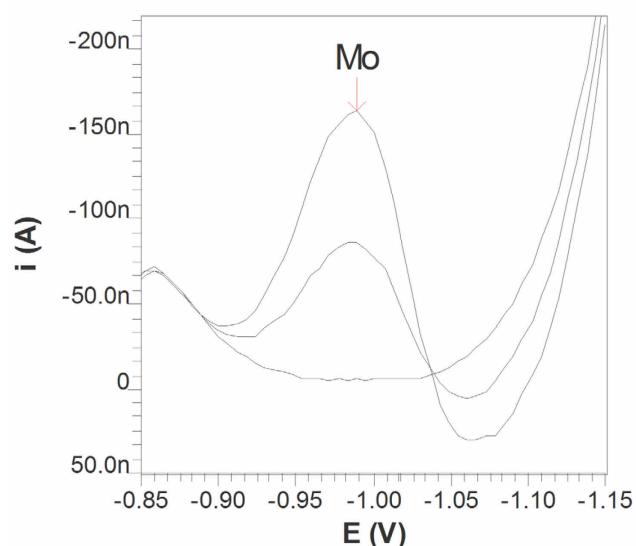
For determinations in which an extremely low chemical blank value is desired, the self-synthesizing 5-sulfo-7-nitro-hydroxyquinoline (monohydrate) can be used instead of the commercially available 8-hydroxy-7-iodo-quinoline-5-sulfonic acid.

Blank value 8-hydroxy-7-iodo-quinoline-5-sulfonic acid

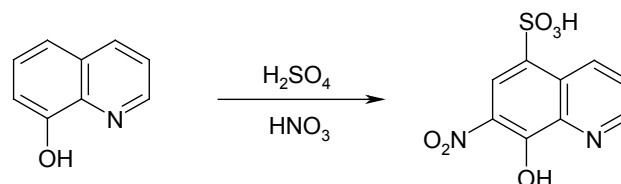


Sample volume	10 mL
Result	22 ng/L Mo

Blank value 7-nitro-8-hydroxyquinoline-5-sulfonic acid



Synthesis of the 7-nitro-8-hydroxyquinoline-5-sulfonic acid monohydrate



- 8-Hydroxyquinoline, CAS 148-24-3
- Oleum (fuming sulphuric acid, 30 % SO₃), CAS 8014-96-7
- Nitric acid, w(HNO₃) = 65 %

Dissolve 4.335 g (0.03 mol) 8-hydroxyquinoline in 50 mL oleum at 50 °C. The solution is cooled to 0 °C and 7 mL nitric acid are added dropwise with vigorous stirring and cooling.

Stirring is continued for 15 min and the solution is then poured onto 450 g crushed ice. The by-product, 5,7-dinitro-8-hydroxyquinoline-5-sulfonic acid monohydrate, is precipitated and filtered off.

The filtrate is stored for two days at -21 °C. After this time, 7-nitro-8-hydroxyquinoline-5-sulfonic acid monohydrate is precipitated. The precipitate is filtered off, washed several times with small amounts of ice water and finally recrystallized in water. It is dried in vacuum at 1 Pa and 23 °C.

Interferences from other ions

Experiments have shown that potential interfering ions (PII) such as Zn²⁺, Co²⁺, Mn²⁺, Ca²⁺, Al³⁺ and TI⁺ cause no interference at molar ratios (PII) / Mo⁶⁺ ≤ 10⁺³.

Tungsten W(VI), on the other hand, exhibits practically the same electrochemical behavior as molybdenum and is hence determined at the same time. With molar ratios W⁶⁺ / Mo⁶⁺ ≤ 10, the masking of tungsten with 0.001 mol/L tartaric acid is complete and there is no interference with the molybdenum determination. With a greater percentage of tungsten, 0.01 mol/L tartaric acid is needed. The sensitivity of the Mo determination is then lowered to app. 37% of the original value (without tartaric acid addition).

References

- Magyar B., Wunderli S.
Microchimica Acta (Wien), 1985/III, 223 – 237
- Stach B., Schöne K.
Microchimica Acta (Wien), 1977/II, 569
- Edmonds T.E.
Anal. Chim. Acta 116(1980), 323 – 333

- Bosserman P., Sawyer D.T., Page A.L.
Anal. Chem. 50(1978), 1300 - 1303
- Lanza P., Ferri D., Buldini P.A.
Analyst 105(1980), 379 - 385
- Metrohm Application Bulletin 132

Appendix

Report for the example determination of a molybdenum determination in drinking water

```
===== METROHM 746 VA TRACE ANALYZER (5.746.0101) =====
Determ.      : 11161153      User: zu      Date: 1999-11-16
Modified     : 1999-11-16 12:02:35 Run : 4      Time: 11:53:35
Sample table: -
```

Pos.	Ident.1/S1 Test Mo	Ident.2/S2	Ident.3/S3	Method.call	Sample size/S0 10 mL
------	-----------------------	------------	------------	-------------	-------------------------

```
Method : AB146
Title  : Determination of Molybdenum in Waters. AB146
Remark1: 10 mL tap water + 2 mL reagent solution
Remark2 :
```

```
Substance : Mo
Mass conc.: 245.6 ng/L      Mass : 2.456 ng
MC.dev.   : 4.12 ng/L (1.68%) Add.mass : 1 ng
Cal.dev.  : -              V0.sample: 10 mL
```

VR	U/mV	I/nA	I.mean	Std.dev.	I.delta	Comments
00	-1042	-396.3	-399.0	3.886		crit. front ovlp.
01	-1042	-401.8				crit. front ovlp.
10	-1041	-561.8	-561.7	0.1205	-162.7	
11	-1041	-561.6				
20	-1040	-711.8	-713.2	1.913	-151.5	
21	-1040	-714.5				

Substance	Techn.	Y.reg/offset	Slope	Nonlin.	Mean deviat.
Mo	std.add.	-4.004e-07	-1.956		3.322e-09

```
C# Workg.com.var Remark
```

Final results	+/-	Res.dev.	%	Comments
Mo = 245.63 ng/L		4.12	1.68	

Method print for the determination of molybdenum

```
===== METROHM 746 VA TRACE ANALYZER (5.746.0101) =====
Method: AB146 .mth OPERATION SEQUENCE
Title : Determination of Molybdenum in Waters. AB146
```

	Instructions	t/s	Main parameters	Auxiliary parameters
1	DOS>M		Soln.name reagent_sol	V.add 2.000 mL
2	SMPL>M		V.fraction mL	V.total L
3	REM		10 mL sample + 2 mL reagent solution	
4	PURGE			
5	STIR	300.0	Rot.speed 2000 /min	
6	(ADD			
7	STIR		Rot.speed 2000 /min	
8	PURGE	30.0		
9	SEGMENT		Segm.name csv	
10	ADD>M		Soln.name Mo-std	V.add 0.100 mL
11	ADD)2			
12	END			

Method: AB146 SEGMENT csv

	Instructions	t/s	Main parameters	Auxiliary parameters
1	0PURGE			
2	0STIR	5.0		
3	(REP			
4	SMDE		Drop size 6	
5	DPMODE		U.ampl -50 mV	t.meas 20.0 ms
			t.step 0.60 s	t.pulse 40.0 ms
6	SWEEP	52.2	U.start -680 mV	U.step 6 mV
			U.end -1180 mV	Sweep rate 10 mV/s
7	OMEAS		U.standby mV	
8	REP)1			
9	PURGE			
10	STIR		Rot.speed 2000 /min	
11	END			