

Application Bulletin 116/4 e

Determination of chromium in small quantities by polarography and adsorptive stripping voltammetry after digestion

Summary

This Application Bulletin describes methods for the polarographic and voltammetric determination of small quantities of chromium in water, effluent water and biological samples. Methods for the sample preparation for various matrices are given.

Instruments

VA instrument capable of operating a Multi-Mode Electrode and supporting differential pulse (DP) measuring mode	
909 UV Digester	2.909.0014

Electrodes

WE	Multi-Mode Electrode pro	6.1246.120
	Mercury drop capillary	6.1226.030
	or	6.1226.050
RE	Ag/AgCl reference electrode	6.0728.x20
	Ag/AgCl/KCl (3 mol/L)	
	Electrolyte vessel Filled with c(KCl) = 3 mol/L	6.1245.010
AE	Pt rod electrode	6.0343.x00

Sample preparation

Reagents

All of the used reagents must be of purest quality possible (for analysis or for trace analysis*).

- Sulfuric acid, for trace analysis*, w(H₂SO₄) = 96%, CAS 7664-93-9
- Hydrogen peroxide solution, for trace analysis*, w(H₂O₂) = 30%, CAS 7722-84-1
- Hydrochloric acid, for trace analysis*, w(HCl) = 30%, CAS 7647-01-0
- Methyl isobutyl ketone (MIBK), for analysis, CAS 108-10-1

- Sodium diethyl dithiocarbamate trihydrate (NaDDTC), CAS 20624-25-3
- Potassium hydrogen phthalate, for analysis, CAS 877-24-7
- Sodium sulfite, for analysis, CAS 7757-83-7
- Potassium peroxydisulfate, for analysis, CAS 7727-21-1
- Potassium permanganate, for analysis, KMnO₄, CAS 7722-64-7
- Sodium hydroxide solution, for trace analysis*, w(NaOH) = 30%, CAS 1310-73-2
- Ultrapure water, resistivity >18 MΩ·cm (25 °C), type I grade (ASTM D1193)
- Water-repellent filter paper, diameter 9 cm

* e.g., Merck suprapur®, Honeywell Fluka TraceSelect® or equivalent

Solutions

Diluted sulfuric acid	c(H ₂ SO ₄) = 1.0 mol/L
Sodium diethyl dithiocarbamate solution	w(NaDDTC) = 2 % in water The solution should always be filtered before use.
Potassium hydrogen phthalate buffer (pH 4.0)	c(potassium hydrogen phthalate) = 0.05 mol/L
Potassium permanganate solution	c(KMnO ₄) = 0.02 mol/L
Diluted sodium hydroxide solution	c(NaOH) = 0.1 mol/L

Sample preparation

- For sulfite-free surface or drinking water no special preparation is necessary. The samples can be analyzed directly as described under «Analysis».
- With slightly contaminated waste water UV digestion has proved to be best. Add 10 µL conc. hydrochloric acid and 50 µL H₂O₂ to 10 mL of the sample solution and treat in the 909 UV Digester during 90 min at 90°C.

- Water containing also sulfite and Cr(III) next to Cr(VI): In acidic sulfite solution, Cr(VI) is reduced to Cr(III). Since one is primarily interested in the Cr(VI) fraction hexavalent chromium must first be extracted and determined separately.
 - Adjust the sample to pH = 4 with diluted sulfuric acid 1 mol/L.
 - Place 50 mL of this solution in a 100 mL separating funnel with 5 mL of pH 4.0 buffer solution, 2 mL NaDDTC and 10 mL MIBK. Shake vigorously for 2 minutes to extract the Cr(VI).
 - Allow to stand for 15 min, drain off the greater portion of the aqueous phase and filter the MIBK portion via a phase-separating filter into a 50 mL Kjeldahl flask. Rinse the separating funnel with a further 3 mL MIBK and pass this through the filter into the Kjeldahl flask. Add 10 mL distilled water, a glass boiling bead and 2 mL conc. sulfuric acid.
 - Place the Kjeldahl flask in a boiling water bath and drive off the MIBK with a strong stream of nitrogen. Rinse the gas inlet tip with distilled water and add 1.8 mL H₂O₂ to the still hot solution.
 - Heat over a Bunsen flame until the H₂O₂ starts to decompose. Stop heating until reaction is completed and the sulfuric acid is finally evaporated down to about 0.5 mL. Digestion is now completed, and the sample may be allowed to cool.
- Biological samples and waste water containing a high proportion of organic matter and chlorides proceed as follows: In the presence of sulfuric acid, Cr(VI) combines with chloride ions to form the volatile complex chromyl chloride (CrO₂Cl₂), whose boiling-point is 117 °C. To avoid loss of chromium owing to evaporation of chromyl chloride, add a spatula-tip of sodium sulfite to the sample before digestion. Digest as described in Application Bulletin 113: «Sample preparation».

Oxidation from Cr(III) to Cr(VI)

Oxidation with KMnO₄

- Add 10 mL distilled water and 2 drops KMnO₄ solution to the cooled sulfuric acid and heat to boiling-point.
- Continue adding KMnO₄ drop by drop until the pink color remains. While keeping the total volume constant by adding small amounts of water, allow the solution to boil approx. 5 min.
- Add diluted NaOH solution to the still hot solution to bring the pH value to 5-9.

- Now the solution can be cooled and rinsed into the polarography vessel with distilled water.

Oxidation with K₂S₂O₈

Another possibility is the oxidation with K₂S₂O₈.

- Add 10 mL distilled water and a spatula-tip of K₂S₂O₈ to the cooled sulfuric acid digestion solution
- Boiling slightly, allow solution to boil down to approx. 0.5 mL, then cool and afterwards fill up to 10 mL.
- The surplus of peroxodisulfate must be boiled off, otherwise traces of peroxide will interfere with the analysis.

Oxidation with UV in the 909 UV Digester at pH 4-6

- To do this, adjust 10 mL of the sample solution to pH 4-6, then add 50 µL H₂O₂. Afterwards expose the sample solution to treatment in the UV Digester at 90°C for 30 min.

Method 1: Polarographic determination of Cr concentrations >10 µg/L

Summary

Higher chromium concentrations are determined at the dropping mercury electrode. Cr(III) must first be wet chemically oxidized. The Cr(IV) content is determined by means of DP polarography.

Reagents

All of the used reagents must be of purest quality possible (for analysis or for trace analysis*).

- Potassium hydroxide, for trace analysis* w(KOH) = 30%, CAS 1310-58-3
- Ammonia solution, for trace analysis*, w(NH₃) = 25%, CAS 1336-21-6
- Acetic acid, for trace analysis*, w(CH₃COOH) = 96 - 100 %, CAS 64-19-7
- Ethylene diamine, for analysis, CAS 107-15-3
- Cr standard stock solution: β(Cr⁶⁺) = 1 g/L (commercially available)

- Ultrapure water, resistivity >18 MΩ·cm (25 °C), type I grade (ASTM D1193)

* e.g., Merck suprapur®, Honeywell Fluka TraceSelect® or equivalent

Solutions

Diluted ammonia solution	w(NH ₃) = 5%
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Standard solutions

Cr(VI) standard solution	β(Cr ⁶⁺) = 1 mg/L
	The diluted standard solutions (e.g. 1 mg/L Cr) are prepared from a standard stock solution by dilution in water. They are freshly prepared daily.

Analysis

Measuring solution

10 mL (diluted) sample or digestion solution

10 µL ethylene diamine

150 µL acetic acid

200 µL diluted ammonia solution

Adjust the pH value of the solution with KOH or acetic acid to pH 6.8 ± 0.1. If necessary, allow to cool. The solution is purged with nitrogen for 600 s.

The chromium concentration is determined by means of the standard addition method.

Parameters

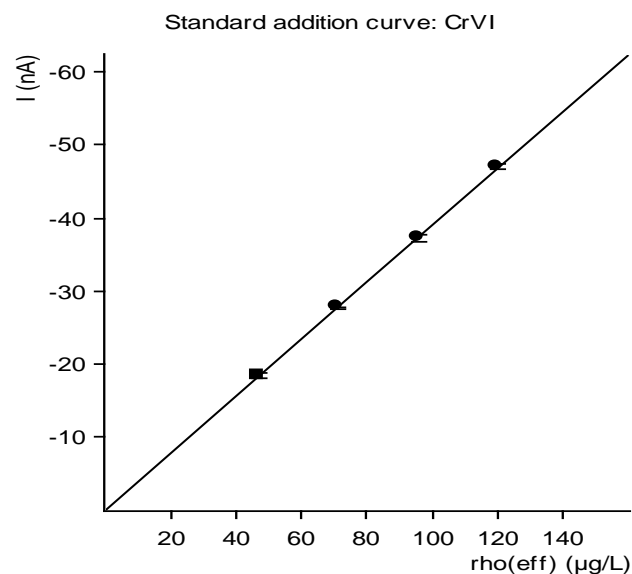
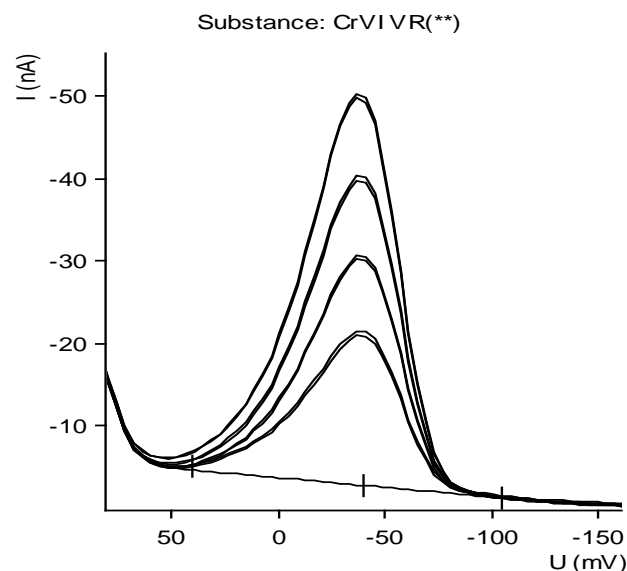
Voltammetric

Electrode operating mode	SMDE
Measuring mode	DP – Differential pulse
Stirring rate	2000 min ⁻¹
Equilibration time	3 s
Sweep	
Start potential	0.10 V
End potential	-0.17 V
Potential step	0.004 V
Potential step time	0.6 s
Sweep rate	0.0067 V/s
Pulse amplitude	0.05 V

Substance

Name	Cr
Characteristic potential	-0.04 V

Example



Result

Sample	Waste water
Sample size	10 mL
β(Cr)	47.7 µg/L

Comments

- The solution has to be degassed for min. 10 min to prevent interferences from oxygen.

- The limit of detection in samples is 10 µg/L, the upper limit of the linear range is 1 mg/L Cr. Samples of higher concentrations must be diluted.
- We recommend an evaluation of the curves using the „linear“ or „exponential“ baseline.
- Weak organically contaminated waste water can be analyzed without digestion.

Methods 2 + 3: Determination of chromium traces using adsorptive stripping voltammetry

Summary

Cr(III) ions or Cr(IV) ions after „in situ“ reduction at the mercury electrode, build a complex with DTPA. In this complex the Cr(III) is reduced to Cr(II) and oxidized again catalytically to Cr(III) by nitrate ions. The current flowing during this process can be used for the quantitative determination of Cr.

Reagents

All of the used reagents must be of purest quality possible (for analysis or for trace analysis*).

- Sodium acetate, for trace analysis*, CAS 127-09-3
- Diethylenetriaminepentaacetic acid (DTPA), Titriplex™ V, for analysis, CAS 67-43-6
- Sodium nitrate, NaNO₃, for trace analysis*, CAS 7631-99-4
- Sodium hydroxide solution, for trace analysis*, w(NaOH) = 30 %, CAS 1310-73-2
- Cr standard stock solution: β(Cr⁶⁺) = 1 g/L (commercially available)
- Ultrapure water, resistivity >18 MΩ·cm (25 °C), type I grade (ASTM D1193)

* e.g., Merck suprapur®, Honeywell Fluka TraceSelect® or equivalent

Solutions

Supporting electrolyte	c(sodium acetate) = 0.2 mol/L c(DTPA) = 0.05 mol/L c(NaNO ₃) = 2.5 mol/L
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1.64 g sodium acetate, 1.96 g DTPA and 21.3 g sodium nitrate are dissolved in ultrapure water and filled up to 100 mL.

Method 2: Cr concentrations between 0.02 and 1.5 µg/kg

Standard solution

Cr(VI) standard solution	β(Cr ⁶⁺) = 0.02 mg/L Diluted standard solutions are prepared from a standard stock solution by dilution in water. They are freshly prepared daily.
--------------------------	---

Analysis

Measuring solution

10 mL (diluted) sample

2.5 mL supporting electrolyte

Adjust the pH value of this solution to 6.2 ± 0.1, using diluted NaOH solution.

The chromium concentration is determined by means of the standard addition method.

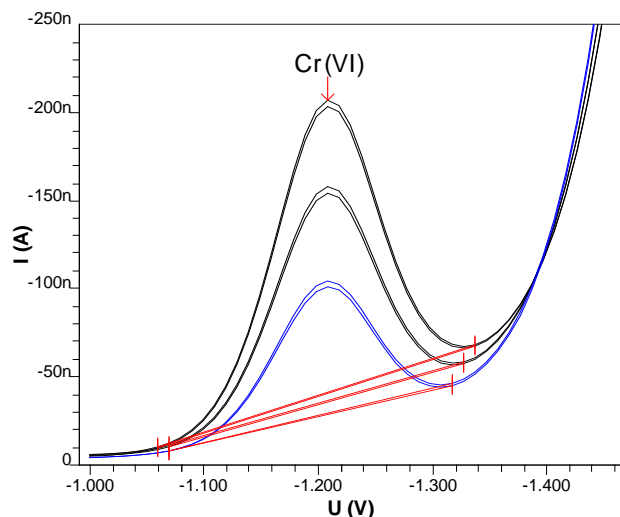
Parameters

Voltammetric

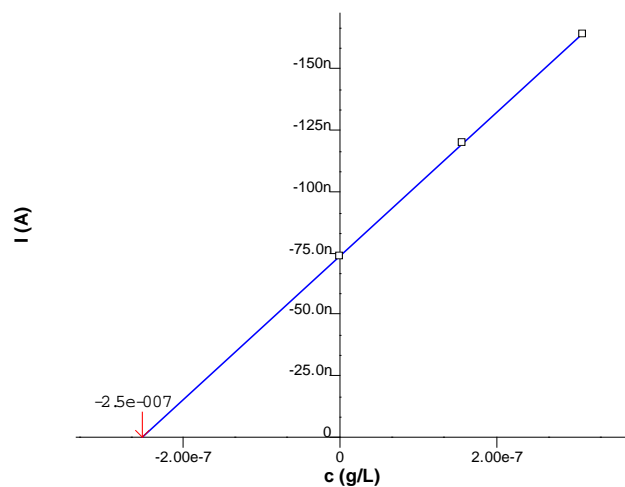
Electrode operating mode	HMDE
Drop size	7
Measuring mode	DP – Differential pulse
Stirring rate	2000 min ⁻¹
<i>Potentiostatic pretreatment</i>	
Potential 1	-1.0 V
Waiting time 1	60 s
Equilibration time	10 s
<i>Sweep</i>	
Start potential	-1.0 V
End potential	-1.45 V
Potential step	0.01 V
Potential step time	0.3 s
Sweep rate	0.033 V/s
Pulse amplitude	0.05 V

Substance

Name	Cr
Characteristic potential	-1.25 V

Example


Cr(VI)
 c = 0.316 ug/l
 +/- 0.010 ug/l (3.26%)


Result

Sample	Tap water
Sample size	10 mL
$\beta(\text{Cr})$	0.32 $\mu\text{g/L}$

Method 3: Cr concentrations between 1 and 5 $\mu\text{g/L}$

Standard addition solution

Cr(VI) standard solution	$\beta(\text{Cr}^{6+}) = 0.25 \text{ mg/L}$ Diluted standard solutions are prepared from a standard stock solution by dilution in water. They are freshly prepared daily.
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Analysis
Measuring solution

10 mL (diluted) sample
 2.5 mL supporting electrolyte

Adjust the pH value of this solution to 6.2 ± 0.1 , using diluted NaOH solution. The chromium concentration is determined by means of the standard addition method.

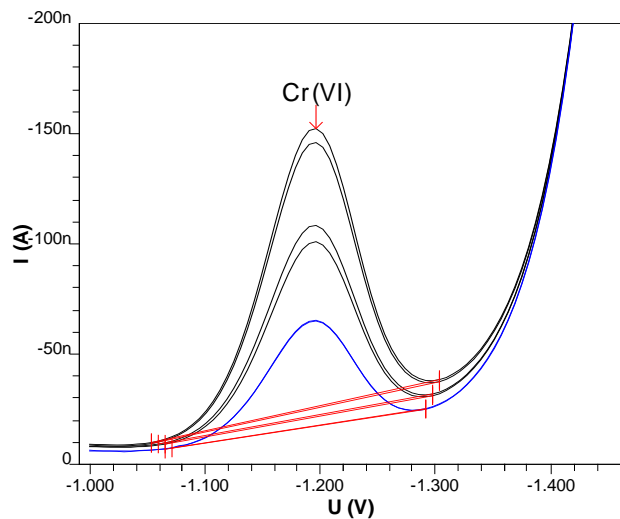
Parameters
Voltammetric

Electrode operating mode	HMDE
Drop size	7
Measuring mode	DP – Differential pulse
Stirring rate	2000 min^{-1}
Equilibration time	10 s
Sweep	
Start potential	-1.0 V
End potential	-1.5 V
Potential step	0.01 V
Potential step time	0.3 s
Sweep rate	0.033 V/s
Pulse amplitude	0.05 V

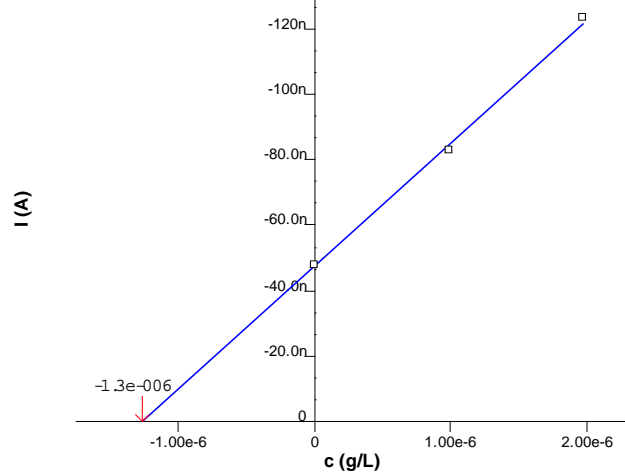
Substance

Name	Cr
Characteristic potential	-1.25 V

Example



Cr(VI)
 c = 1.585 µg/l
 +/- 0.110 µg/l (6.92%)



Result

Sample	Sea water
Sample size	10 mL
$\beta(\text{Cr})$	1.6 µg/L

Comments

- For methods 2 and 3 chromium should be present as Cr(VI), since with pure Cr(III) solutions peak heights constantly diminish. Sensitivity with the Cr(VI) solutions is far greater than with the Cr(III) solutions.
- Higher Mg concentrations (>100 mg/L) interfere with the Cr(VI) determination. The background current greatly increases. It is recommended to dilute the sample solutions, e.g. seawater.

References

- Golimowski J., Valenta P., Nürnberg H. W. Trace determination of chromium in various water types by adsorption differential pulse voltammetry *Fresenius Z. Anal. Chem.* 322, (1985) 315-322
- Scholz F., Lange B., Draheim M., Pelzer J. The catalytic adsorptive stripping voltammetric determination of chromium with DTPA and nitrate *Fresenius Z. Anal. Chem.* 388, (1990) 627-629

Appendix

Report for the example determination of chromium according to method 1

```
===== METROHM 746 VA TRACE ANALYZER (5.746.0101) =====
Determin.      : 05281203          User:          Date: 1999-05-28
Modified       : 2000-12-04 11:59:21 Run : 0          Time: 12:03:14
Sample table:  -
```

```
-----
Pos.  Ident.1/S1  Ident.2/S2  Ident.3/S3  Method.call  Sample size/S0
-----
      waste water
-----
Method : AB116_1
Title  : Chromium Determination. AB116 Part 1
Remark1 : 10 mL sample + 10 µL ethylene diamine + 150 µL acetic acid
Remark2 : + 200 µL NH3 + NaOH --> pH 6.8
-----
```

Substance	CrVI	Mass conc.	47.68 ug/L	Mass	476.8 ng	Comments
MC.dev.	0.952 ug/L (2%)	Add.mass	250 ng			
Cal.dev.	-	V0.sample	10 mL			

VR	U/mV	I/nA	I.mean	Std.dev.	I.delta	Comments
00	-40	-18.17	-18.36	0.2621		
01	-40	-18.54				
10	-40	-27.49	-27.57	0.1061	-9.210	
11	-40	-27.64				
20	-40	-36.77	-37.07	0.4113	-9.499	
21	-40	-37.36				
30	-39	-46.40	-46.67	0.3835	-9.605	
31	-39	-46.94				

Substance	Techn.	Y.reg/offset	Slope	Nonlin.	Mean deviat.
CrVI	std.add.	-1.823e-08	-3.960e-04		2.948e-10

C# Workg.com.var Remark
 -- -----

Final results	CrVI =	47.683 ug/L	+/-	Res.dev.	0.952	%	2.00	Comments

Method print for the determination of chromium according to method 1

```
===== METROHM 746 VA TRACE ANALYZER (5.746.0101) =====
Method: AB116_1.mth          OPERATION SEQUENCE
Title : Chromium Determination. AB116 Part 1
-----
```

	Instructions	t/s	Main parameters	Auxiliary parameters
1	DOS/M		V.added 0.360 mL	
2	SMPL/M		V.fraction mL	V.total L
3	STIR		Rot.speed 2000 /min	
4	PURGE	600.0		
5	(ADD			
6	PURGE			
7	STIR	10.0	Rot.speed 2000 /min	
8	SEGMENT		Segm.name pol	
9	ADD>M		Soln.name Cr-Std	V.add 0.050 mL
10	ADD)3			
11	END			

```
Method: AB116_1          SEGMENT
                        pol
-----
```

	Instructions	t/s	Main parameters	Auxiliary parameters
1	0PURGE			
2	0STIR	3.0		
3	(REP			
4	SMDE		Drop size 4	
5	DPMODE		U.ampl -50 mV	t.meas 20.0 ms
			t.step 0.60 s	t.pulse 40.0 ms
6	SWEEP	42.6	U.start 100 mV	U.step 4 mV
			U.end -170 mV	Sweep rate 6.667 mV/s

```

7      OMEAS          U.standby          mV
8      REP)1
9      PURGE
10     STIR          Rot.speed          2000 /min
11     END
  
```

```

Method: AB116_1          SUBSTANCES
                        CrVI          - pol
  
```

```

-----
Recognition              Display / Plot
-----
U.verify                 -40 mV          I.scale          auto
U.tol (+/-)             30 mV          U.div            50.00 mV/cm
U.width min              10 mV          U.begin          80 mV
U.width max              200 mV         U.end            -160 mV
I.threshold              200 pA
  
```

```

-----
Baseline                 Evaluation
-----
Type                     linear          Mode              VA
Scope                    whole           Quantity          I.peak
dU.front                 auto           Sign. digits      4
S.front                  auto
dU.rear                  auto
S.rear                   auto
  
```

```

-----
Calibration              1999-05-28 12:17:00      Coefficients
-----
Technique                std.add.              Y.reg            -1.823e-08
Curve type               linear                 Slope            -0.0003892
                        Nonlin.
                        Mean dev.            2.95e-10
  
```

Additions

```

-----
Soln.name                Cr-Std
-----
Mass conc.               10 mg/L             g/L              g/L              g/L
Range min                g/L                  g/L              g/L
Range max                g/L                  g/L              g/L
M.conc./cm               g/L                  g/L              g/L
  
```

```

Method: AB116_1          CALCULATION
                        max. 15 lines
  
```

```

-----
Quantity                 Formula (R##, C##, A##)      Res.unit         Sig.dig.
-----
CrVI                     R1000=MC:CrVI              #g/L             5
  
```

Report for the example determination of chromium according to method 2

```

===== METROHM 797 VA COMPUTRACE (Version 1.0.0.1) (Serial No. 0) =====
  
```

```

Determination : 11271032_tap water.dth
Sample ID     : tap water
Creator method :          Date :          Time:
Creator determ.:          Date : 2000-11-27      Time: 10:32:35
Modified by   : ---       Date :          Time:
  
```

```

-----
Method          : AB 116 M2.mth
Title           : AB 116/3 Determination of Cr (Method 2)
Remark1        : 10 mL sample + 2.5 mL supporting electrolyte
Remark2        :
  
```

```

-----
Sample amount   : 10.000 mL
Cell volume     : 12.500 mL
  
```

```

-----
Substance      : Cr(VI)
Conc.          : 252.741 ng/L
Conc.dev.      : 8.247 ng/L      ( 3.26%)
Amount         : 3.159 ng
Add.amount     : 2.000 ng
  
```

```

-----
VR      V      nA      I.mean  Std.Dev.  I.delta  Comments
-----
1 - 1   -1.208  -74.9    -73.7    1.795    0.0
1 - 2   -1.208  -72.4
2 - 1   -1.208  -121.3   -119.9   2.053    -46.2
2 - 2   -1.208  -118.4
3 - 1   -1.208  -165.6   -164.1   2.045    -44.3
  
```


3 - 2 -1.208 -162.7

Substance	Calibr.	Y.reg/offset	Slope	Mean deviat.	Corr.Coeff.
Cr(VI)	std.add.	-7.375e-008	-2.918e-001	2.362e-009	0.99929

Final results	+/-	Res. dev.	%	Comments
Cr(VI): default	=	0.316 ug/l	0.010	3.263

Method print for the determination of chromium according to method 2

Method parameters

Method : AB116_2_Det of CrVI with HMDE.mth
 Title : Determination of Chromium(VI). AB116 part 2
 Remark1 : 10ml sample + 2.5ml buffer --> pH 6.2 ± 0.1 with NaOH
 Remark2 : buffer.: 0.2mol/l sodium acetate + 0.05mol/l DTPA + 2.5 mol/l NaNO3

Calibration : Standard addition
 Technique : Batch
 Addition : Manual

Sample ID : oxydized sample
 Sample amount (mL): 10.000
 Cell volume (mL): 12.500

Voltammetric parameters

Mode : DP - Differential Pulse

Highest current range : 10 mA
 Lowest current range : 100 nA

Electrode : HMDE
 Drop size (l..9) : 4
 Stirrer speed (rpm) : 2000

Initial electr. conditioning : No

No. of additions : 2
 No. of replications : 2

Measure blank : No
 Addition purge time (s) : 10

Initial purge time (s) : 300

Conditioning cycles
 Start potential (V) : 0.000
 End potential (V) : 0.000
 No. of cycles : 0

Hydrodynamic (measurement) : No
 Cleaning potential (V) : 0.000
 Cleaning time (s) : 0.000
 Deposition potential (V) : -1.000
 Deposition time (s) : 60.000

Sweep
 Equilibration time (s) : 10.000
 Start potential (V) : -1.000
 End potential (V) : -1.450
 Voltage step (V) : 0.010
 Voltage step time (s) : 0.300
 Sweep rate (V/s) : 0.033
 Pulse amplitude (V) : 0.050
 Pulse time (s) : 0.040

Cell off after measurement : Yes

Peak evaluation

Regression technique : Linear Regression
 Peak evaluation : Height
 Minimum peak width (V.steps) : 5
 Minimum peak height (A) : 1.000e-010
 Reverse peaks : No

Smooth factor : 4
 Eliminate spikes : Yes

Substances

 Cr(VI) : -1.250 V +/- 0.050 V
 Standard solution : 1 20.000 ug/L
 Addition volume (mL) : 0.100
 default : Final result (Cr(VI)) =
 Conc * (12.5 / 10) * (1e+006 / 1) + 0 - 0

Baseline

Substance	Addition	automatic	start (V)	end (V)	type	scope
Cr(VI)	Sample	yes	---	---	linear	wholePeak
	Addition 1	yes	---	---	linear	wholePeak
	Addition 2	yes	---	---	linear	wholePeak

Report for the example determination of chromium according to method 3

===== METROHM 797 VA COMPUTRACE (Version 1.0.0.1) (Serial No. 0) =====
 Determination : 11271458_sea water.dth
 Sample ID :
 Creator method : Date : Time:
 Creator determ.: Date : 2000-11-27 Time: 14:58:48
 Modified by : Date : 2017-07-10 Time: 13:37:11

 Method : AB 116 M3.mth
 Title :
 Remark1 : 10 mL sample + 2.5 mL supporting electrolyte
 Remark2 :

 Sample amount : 10.000 mL
 Cell volume : 12.500 mL

 Substance : Cr(VI)
 Conc. : 1.268 ug/L
 Conc.dev. : 0.088 ug/L (6.92%)
 Amount : 15.849 ng
 Add.amount : 12.500 ng

VR	V	nA	I.mean	Std.Dev.	I.delta	Comments
1 - 1	-1.196	-47.5	-47.8	0.372	0.0	
1 - 2	-1.196	-48.0				
2 - 1	-1.196	-79.5	-82.8	4.691	-35.1	
2 - 2	-1.190	-86.1				
3 - 1	-1.196	-126.1	-123.3	3.956	-40.5	
3 - 2	-1.196	-120.5				

Substance	Calibr.	Y.reg/offset	Slope	Mean deviat.	Corr.Coeff.
Cr(VI)	std.add.	-4.746e-008	-3.743e-002	3.655e-009	0.99582

Final results	+/-	Res. dev.	%	Comments
Cr(VI): default	=	1.585 ug/l	0.110	6.923

Method print for the determination of chromium according to method 3

Method parameters

 Method : AB116_3_Det of CrVI with HMDE.mth
 Title : Determination of Chromium(VI). AB116 part 3
 Remark1 : 10ml sample + 2.5ml buffer --> pH 6.2 ± 0.1 with NaOH
 Remark2 : buffer.: 0.2mol/l sodium acetate + 0.05mol/l DTPA + 2.5 mol/l NaNO3
 Calibration : Standard addition
 Technique : Batch
 Addition : Manual

Sample ID : oxydized sample
Sample amount (mL): 10.000
Cell volume (mL): 12.500

Voltammetric parameters

```

-----
Mode : DP - Differential Pulse

Highest current range : 10 mA
Lowest current range : 100 nA

Electrode : HMDE
Drop size (1..9) : 4
Stirrer speed (rpm) : 2000

Initial electr. conditioning : No

No. of additions : 2
No. of replications : 2

Measure blank : No
Addition purge time (s) : 10

Initial purge time (s) : 300

Conditioning cycles
Start potential (V) : 0.000
End potential (V) : 0.000
No. of cycles : 0

Hydrodynamic (measurement) : No
Cleaning potential (V) : 0.000
Cleaning time (s) : 0.000
Deposition potential (V) : -1.000
Deposition time (s) : 0.000

Sweep
Equilibration time (s) : 10.000
Start potential (V) : -1.000
End potential (V) : -1.450
Voltage step (V) : 0.010
Voltage step time (s) : 0.300
Sweep rate (V/s) : 0.033
Pulse amplitude (V) : 0.050
Pulse time (s) : 0.040

Cell off after measurement : Yes
  
```

Peak evaluation

```

-----
Regression technique : Linear Regression
Peak evaluation : Height
Minimum peak width (V.steps) : 5
Minimum peak height (A) : 1.000e-010
Reverse peaks : No
Smooth factor : 4
Eliminate spikes : Yes
  
```

Substances

```

-----
Cr(VI) : -1.250 V +/- 0.050 V

Standard solution : 1 1.000 mg/L
Addition volume (mL) : 0.100

default : Final result (Cr(VI)) =
          Conc * (12.5 / 10) * (1e+006 / 1) + 0 - 0
  
```

Baseline

```

-----
Substance Addition automatic start (V) end (V) type scope
-----
Cr(VI) Sample yes --- --- linear wholePeak
      Addition 1 yes --- --- linear wholePeak
      Addition 2 yes --- --- linear wholePeak
-----
  
```