

Application Bulletin 74/4 e

Determination of antimony, bismuth, and copper by anodic stripping voltammetry

Summary

This Application Bulletin describes the voltammetric determination of the elements antimony, bismuth, and copper.

The limit of detection for the three elements is 0.5 ... 1 µg/L.

Instruments

VA instrument	capable of operating a Multi-Mode Electrode and supporting differential pulse (DP) measuring mode
909 UV Digester	2.909.0014

Electrodes

WE	Multi-Mode Electrode pro Mercury drop capillary	6.1246.120 6.1226.030 or 6.1226.050
RE	Ag/AgCl reference electrode Ag/AgCl/KCl (3 mol/L) Electrolyte vessel Filled with c(KCl) = 3 mol/L	6.0728.x20 6.1245.010
AE	Pt rod electrode	6.0343.x00

Reagents

All of the used reagents must be of purest quality possible (for analysis or for trace analysis*).

- Hydrochloric acid, for trace analysis*, w(HCl) = 30%, CAS 7647-01-0
- Antimony standard stock solution: $\beta(\text{Sb}^{3+}) = 1 \text{ g/L}$
Commercially available.
- Bismuth standard stock solution: $\beta(\text{Bi}^{3+}) = 1 \text{ g/L}$
Commercially available.
- Copper standard stock solution: $\beta(\text{Cu}^{2+}) = 1 \text{ g/L}$
Commercially available.
- Ultrapure water, resistivity >18 MΩ·cm (25 °C), type I grade (ASTM D1193)

* e.g., Merck suprapur®, Honeywell Fluka TraceSelect® or equivalent

Standard solutions

Sb standard solution	$\beta(\text{Sb}^{3+}) = 1 \text{ mg/L}$ The solution is diluted with c(HCl) = 0.1 mol/L. It is stable for max. 1 week.
Bi standard solution	$\beta(\text{Bi}^{3+}) = 1 \text{ mg/L}$ The solution is diluted with c(HCl) = 0.1 mol/L. It is stable for max. 1 week.
Cu standard solution	$\beta(\text{Cu}^{2+}) = 1 \text{ mg/L}$ The solution is diluted with c(HCl) = 0.1 mol/L. It is stable for max. 1 week.

Sample preparation

- Ground water, surface waters, mineral waters and drinking waters can usually be analyzed without pre-treatment.
- Organic matter often interferes with voltammetric determinations and therefore sample solutions usually have to be digested.
- Low polluted waste waters can be digested with the 909 UV Digester:
- Add 50 - 100 µL w(H₂O₂) = 30% and 10 µL w(HCl) = 30% to 10 mL acidified sample (pH = 2) and irradiate for 1 h at 90 °C. After cooling to room temperature, the digested sample can be transferred directly to the polarographic vessel.
- Samples with organic matter (foods, pharmaceuticals etc.) must be digested.
 - High-pressure ashing
 - Microwave digestion

Both techniques oxidize the sample in a closed digestion vessel by means of a mixture of concentrated mineral acids.

 - Open wet digestion with H₂SO₄ and H₂O₂ according to Application Bulletin 113.

Method 1: Determination of Bi and Sb(III) or Cu

Bi and Sb(III) can be determined simultaneously in $c(\text{HCl}) = 0.6 \text{ mol/L}$. In this electrolyte Cu interferes with the Sb determination since their peak potentials are very close together. The interference can be eliminated by selective oxidation of the Cu after the deposition, prior to the sweep.

Analysis

Measuring solution

10 mL (diluted) sample

0.6 mL hydrochloric acid

The concentration is determined by standard addition.

Parameters

Voltammetric

Electrode operating mode	HMDE
--------------------------	------

Measuring mode	DP – Differential pulse
----------------	-------------------------

Stirring rate	2000 min^{-1}
---------------	-------------------------

Potentiostatic pretreatment

Potential 1	-0.24 V
-------------	---------

Waiting time 1	180 s
----------------	-------

Potential 2	-0.15 V
-------------	---------

Waiting time 2	20 s
----------------	------

Equilibration time	10 s
--------------------	------

Sweep

Start potential	-0.3 V
-----------------	--------

End potential	0.05 V
---------------	--------

Potential step	0.004 V
----------------	---------

Potential step time	0.2 s
---------------------	-------

Sweep rate	0.02 V/s
------------	----------

Pulse amplitude	0.01 V
-----------------	--------

Substance

Name	Bi
------	----

Characteristic potential	-0.01 V
--------------------------	---------

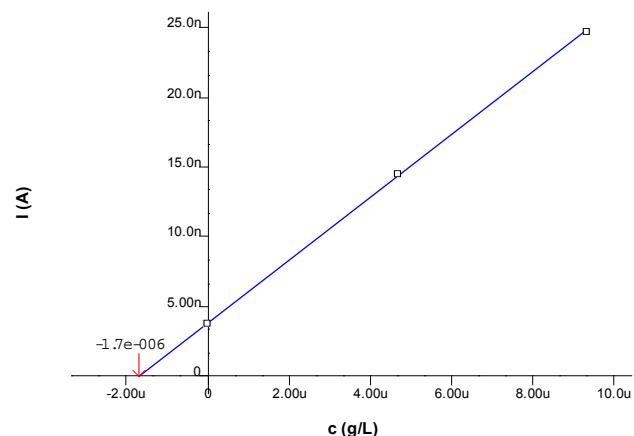
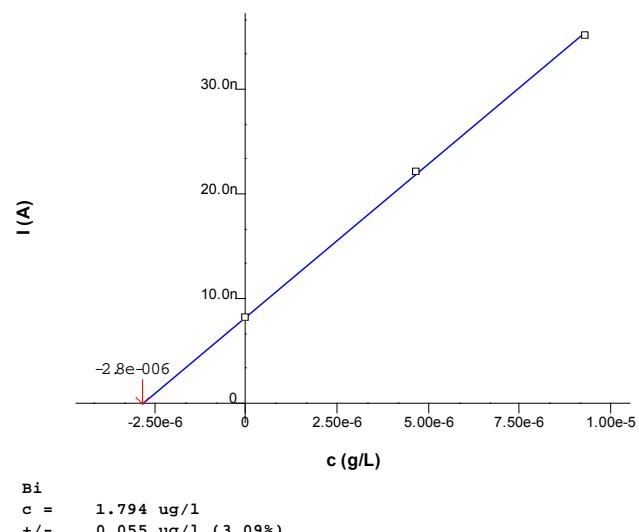
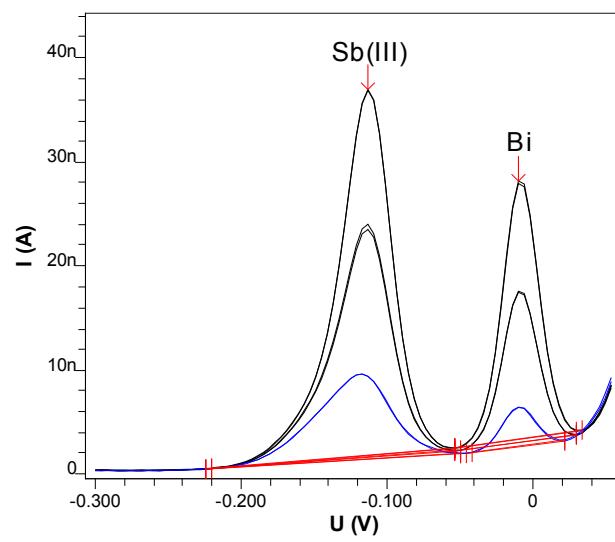
Name	Sb(III)
------	---------

Characteristic potential	-0.11 V
--------------------------	---------

Name	Cu
------	----

Characteristic potential	-0.15 V
--------------------------	---------

Example



Result

Sample	Tap water
Sample size	10 mL
$\beta(\text{Sb}^{\text{III}})$	3.0 $\mu\text{g/L}$
$\beta(\text{Bi})$	1.8 $\mu\text{g/L}$

Comments

- The measuring vessel should contain between 10 ng and 1 μg Sb and Bi each.
- To obtain a better peak shape it is recommended to apply a pulse amplitude of only 10 mV.
- In $\text{c}(\text{HCl}) = 0.6 \text{ mol/L}$ only Sb(III) can be determined. Sb(V) must first be reduced to Sb(III) in this solution. (Evaporate to dryness with sufficient hydrazine sulfate to cover the tip of a spatula). In this way it is also possible to speciate between both oxidation states.
- In $\text{w}(\text{HCl}) = 10\%$ the sum Sb(III) + Sb(V) is determined (refer to method 2).

Method 2: Determination of Cu besides Sb or Bi

In $\text{w}(\text{HCl}) = 10\%$ Cu can be determined simultaneously with Sb(total) or Bi. A separation of Sb and Bi in this electrolyte is not possible since both elements show a peak at the same potential.

Analysis

Measuring solution

10 mL (diluted) sample

5 mL hydrochloric acid

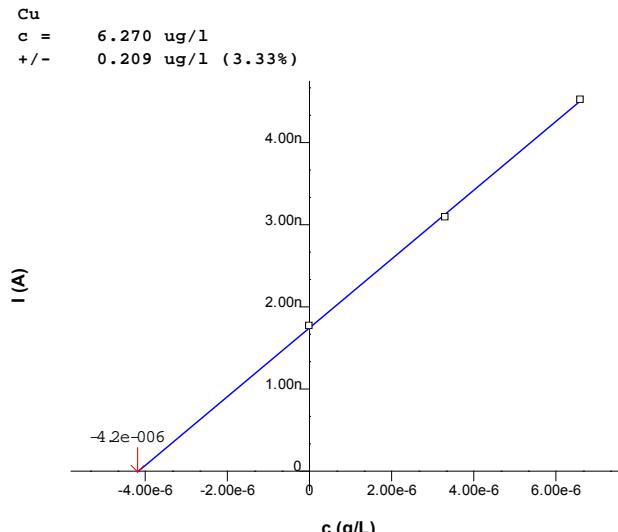
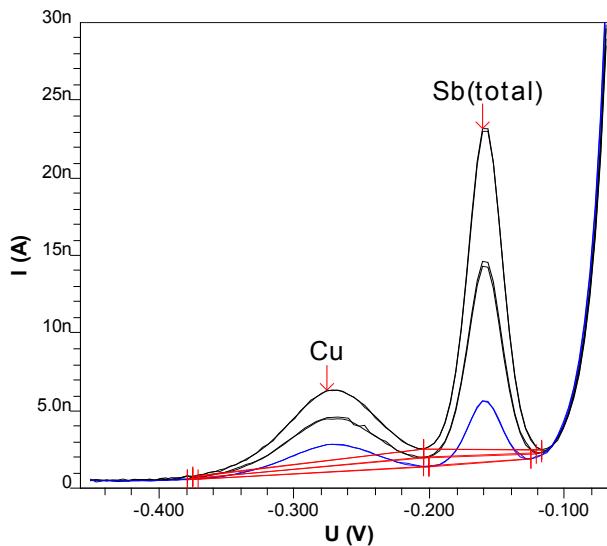
The concentration is determined by two standard additions.

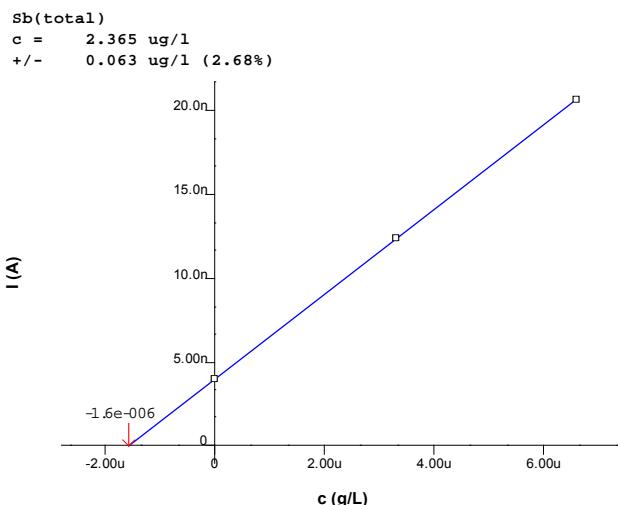
Parameters

Voltammetric	
Electrode operating mode	HMDE
Measuring mode	DP – Differential pulse
Stirring rate	2000 min^{-1}
Potentiostatic pretreatment	
Potential 1	-0.4 V
Waiting time 1	180 s

Equilibration time	10 s
Sweep	
Start potential	-0.45 V
End potential	-0.07 V
Potential step	0.004 V
Potential step time	0.2 s
Sweep rate	0.02 V/s
Pulse amplitude	0.01 V
Substance	
Name	Cu
Characteristic potential	-0.27 V
Name	Sb(total)
Characteristic potential	-0.16 V
Name	Bi
Characteristic potential	-0.16 V

Example





Result

Sample	Tap water
Sample size	10 mL
$\beta(\text{Cu})$	6.3 µg/L
$\beta(\text{Sb}^{\text{total}})$	2.4 µg/L

Comments

- In HCl 10% the sum Sb(III) + Sb(V) is determined with the same sensitivity for both species.

Method 3: Determination of Sb, Bi and Cu

The combination of method 1 and 2 allows the determination of all three elements. First antimony and bismuth are determined in $c(\text{HCl}) = 0.6 \text{ mol/L}$. After further addition of hydrochloric acid also the copper is determined.

	Method 1	Method 2
	$c(\text{HCl}) = 0.6 \text{ mol/L}$	$w(\text{HCl}) = 10 \%$
Sb	-110 mV	-160 mV
Bi	-10 mV	-160 mV
Cu	-150 mV	-270 mV

Analysis

Measuring solution for the determination of Sb and Bi

10 mL (diluted) sample solution

0.6 mL hydrochloric acid

The Sb and Bi concentrations are determined by 2 standard additions each.

Measuring solution for the determination of Cu

Measuring solution from Sb and Bi determination

+ 5 mL hydrochloric acid

The Cu concentration is determined by 2 standard additions.

Parameters

For the determination of Sb and Bi

Voltammetric

Electrode operating mode HMDE

Measuring mode DP – Differential pulse

Stirring rate 2000 min^{-1}

Potentiostatic pretreatment

Potential 1 -0.24 V

Waiting time 1 180 s

Potential 2 -0.15 V

Waiting time 2 20 s

Equilibration time 10 s

Sweep

Start potential -0.3 V

End potential 0.05 V

Potential step 0.004 V

Potential step time 0.2 s

Sweep rate 0.02 V/s

Pulse amplitude 0.01 V

Substance

Name Bi

Characteristic potential -0.01 V

Name Sb(III)

Characteristic potential -0.11 V

For the determination of Cu

Voltammetric

Electrode operating mode HMDE

Measuring mode DP – Differential pulse

Stirring rate 2000 min^{-1}

Potentiostatic pretreatment

Potential 1 -0.4 V

Waiting time 1 180 s

Equilibration time 10 s

Sweep

Start potential -0.45 V

End potential -0.07 V

Potential step 0.004 V

Potential step time 0.2 s

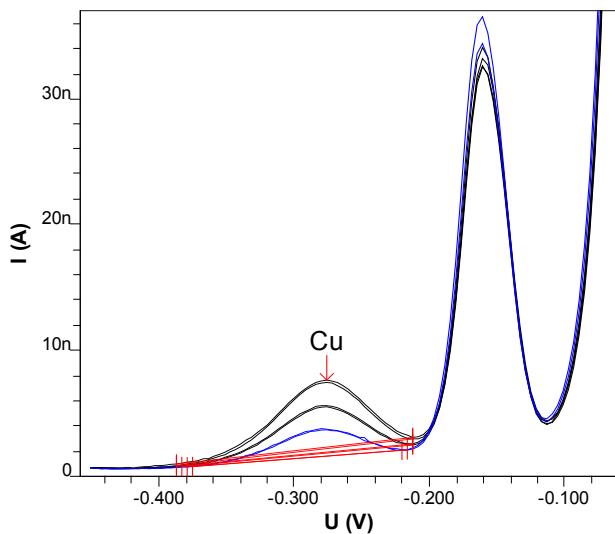
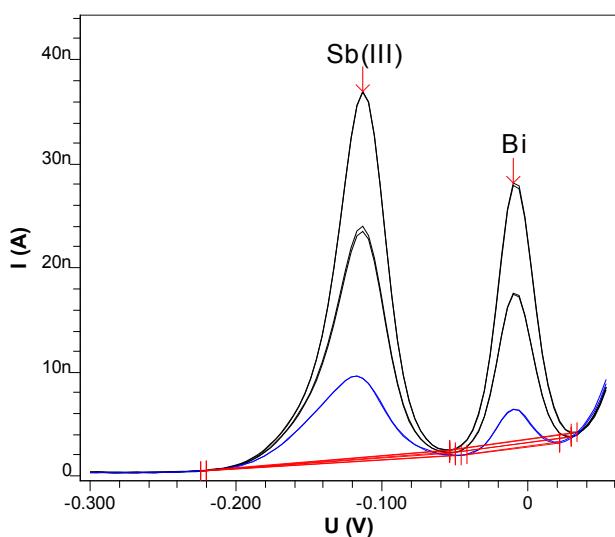
Sweep rate 0.02 V/s

Pulse amplitude 0.01 V

Substance

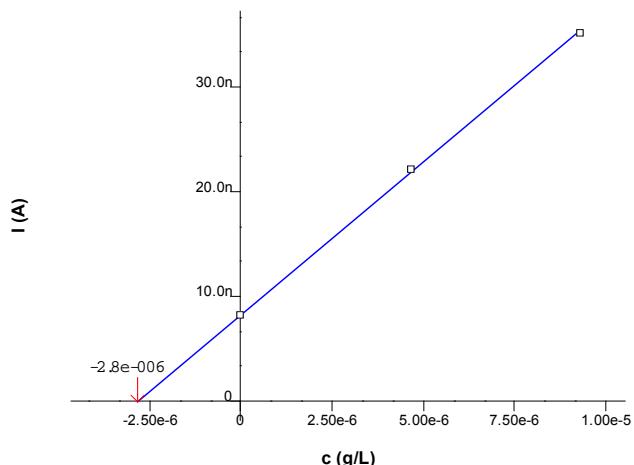
Name Cu

Characteristic potential -0.27 V

Example**Sb(III)**

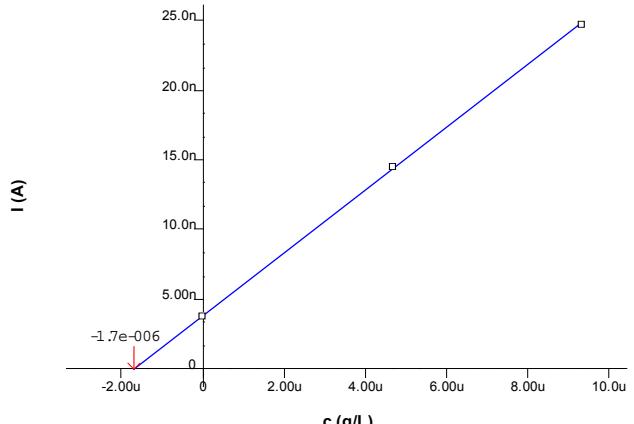
c = 2.998 ug/l

+/- 0.077 ug/l (2.57%)

**Bi**

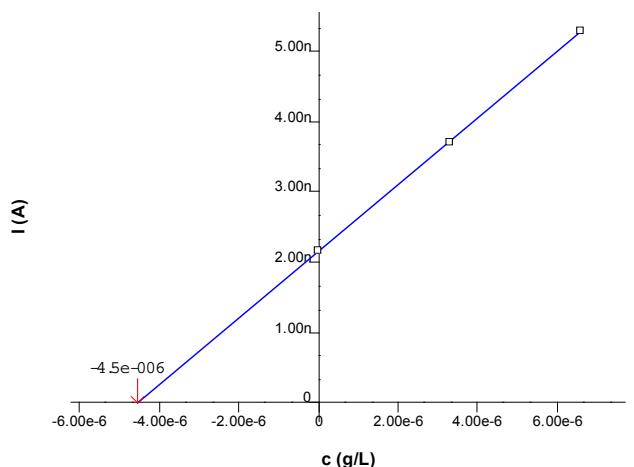
c = 1.794 ug/l

+/- 0.055 ug/l (3.09%)

**Cu**

c = 6.812 ug/l

+/- 0.231 ug/l (3.39%)



Result

Sample	Tap water
Sample size	10 mL
$\beta(\text{Sb}^{\text{III}})$	3.0 $\mu\text{g/L}$
$\beta(\text{Bi})$	1.8 $\mu\text{g/L}$
$\beta(\text{Cu})$	6.8 $\mu\text{g/L}$

Comments

- In 0.6 mol/L HCl only Sb(III) can be determined. Sb(V) must first be reduced to Sb(III) in this solution.
(Evaporate to dryness with sufficient hydrazine sulfate to cover the tip of a spatula). In this way it is also possible to speciate between both oxidation states.
- By combination of method 1 and 2 it is also possible to speciate between both oxidation states of Sb. In c(HCl) = 0.6 mol/L only Sb(III) is determined. In w(HCl) = 10 % total antimony (Sb(III) + Sb(V)) is determined. That is only possible if Bi is absent.

Appendix

Report of the example determination of Sb and Bi according to method 1

```
===== METROHM 797 VA COMPUTRACE (Version 1.0.0.1) (Serial No. 0) =====
Determination : 08161516_SbBi tap water.dth
Sample ID    : SbBi tap water
Creator method:                               Date : Time:
Creator determ.:                            Date : 2000-08-16 Time: 15:16:45
Modified by   :                               Date : 2017-07-13 Time: 13:36:30
-----
Method       : AB74_2_Det of Sb Bi.mth
Title        : Determination of Antimony and Bismuth. AB 74 part 2
Remark1      : 10 ml water + 0.6 ml HCl (30%)
Remark2      :
-----
Sample amount : 10.000 mL
Cell volume   : 10.600 mL
-----
Substance    : Sb(III)
Conc.        : 2.828 ug/L
Conc.dev.    : 0.073 ug/L ( 2.57%)
Amount       : 29.981 ng
Add.amount   : 50.000 ng
-----
VR          V      nA      I.mean  Std.Dev.  I.delta  Comments
----  ----  -----  -----  -----  -----  -----
1 - 1    -0.117  8.22    8.20    0.031    0.00
1 - 2    -0.117  8.18
2 - 1    -0.114  22.31   22.09   0.311    13.89
2 - 2    -0.114  21.87
3 - 1    -0.114  35.13   35.07   0.080    12.98
3 - 2    -0.114  35.02
-----
Substance    : Bi
Conc.        : 1.692 ug/L
Conc.dev.    : 0.052 ug/L ( 3.09%)
Amount       : 17.935 ng
Add.amount   : 50.000 ng
-----
VR          V      nA      I.mean  Std.Dev.  I.delta  Comments
----  ----  -----  -----  -----  -----  -----
1 - 1    -0.010  3.78    3.76    0.031    0.00
1 - 2    -0.010  3.74
2 - 1    -0.010  14.53   14.47   0.083    10.71
2 - 2    -0.010  14.41
3 - 1    -0.010  24.57   24.66   0.131    10.19
3 - 2    -0.010  24.75
-----
Substance    Calibr.      Y.reg/offset      Slope  Mean deviat.  Corr.Coeff.
-----  -----  -----  -----  -----  -----
Sb(III)     std.add.    8.249e-009  2.917e-003  3.422e-010  0.99979
Bi          std.add.    3.814e-009  2.254e-003  1.960e-010  0.99991
-----
Final results           +/- Res. dev. %  Comments
-----  -----  -----  -----
Sb(III):
default      = 2.998 ug/l  0.077    2.572
Bi:
default      = 1.794 ug/l  0.055    3.093
```

Method print for the determination of Sb and Bi according to method 1

```
-----
Method parameters
-----
Method       : AB074_2a_Det of Sb Bi.mth
Title        : Determination of Antimony and Bismuth. AB 74 Method 1
Remark1      : 10 ml water + 0.6 ml HCl (30%)
Remark2      :
-----
Calibration  : Standard addition
Technique    : Batch
Addition    : Manual
-----
Sample ID    : Sample
Sample amount (mL): 10.000
Cell volume  (mL): 10.600
```

Voltammetric parameters

Mode : DP - Differential Pulse

 Highest current range : 10 mA
 Lowest current range : 100 nA

 Electrode : HMDE
 Drop size (1..9) : 4
 Stirrer speed (rpm) : 2000

Initial electr. conditioning : No

 No. of additions : 2
 No. of replications : 2

 Measure blank : No
 Addition purge time (s) : 20

Initial purge time (s) : 300

Conditioning cycles

 Start potential (V) : 0.000
 End potential (V) : 0.000
 No. of cycles : 0

 Hydrodynamic (measurement) : No
 Cleaning potential (V) : -0.240
 Cleaning time (s) : 180.000
 Deposition potential (V) : -0.150
 Deposition time (s) : 20.000

 Sweep
 Equilibration time (s) : 10.000
 Start potential (V) : -0.300
 End potential (V) : 0.050
 Voltage step (V) : 0.004
 Voltage step time (s) : 0.200
 Sweep rate (V/s) : 0.020
 Pulse amplitude (V) : 0.010
 Pulse time (s) : 0.040

Cell off after measurement : Yes

Peak evaluation

 Regression technique : Linear Regression
 Peak evaluation : Height
 Minimum peak width (V.steps) : 10
 Minimum peak height (A) : 1.000e-010
 Reverse peaks : No
 Smooth factor : 4
 Eliminate spikes : Yes

Substances

Sb : -0.110 V +/- 0.050 V

 Standard solution : 1 1.000 mg/L
 Addition volume (mL) : 0.050

 Antimony : Final result (Sb) =
 Conc * (10.6 / 10) * (1e+006 / 1) + 0 - 0

Bi : -0.010 V +/- 0.050 V

 Standard solution : 2 1.000 mg/L
 Addition volume (mL) : 0.050

 Bismuth : Final result (Bi) =
 Conc * (10.6 / 10) * (1e+006 / 1) + 0 - 0

Baseline

Substance Addition automatic start (V) end (V) type scope

 Sb Sample yes --- --- linear wholePeak
 Addition 1 yes --- --- linear wholePeak
 Addition 2 yes --- --- linear wholePeak

 Bi Sample yes --- --- linear wholePeak
 Addition 1 yes --- --- linear wholePeak
 Addition 2 yes --- --- linear wholePeak

Report of the example determination of Sb and Cu in tap water according to method 2

===== METROHM 797 VA COMPUTRACE (Version 1.0.0.1) (Serial No. 0) =====

Determination : 08141708_CuSb 10mV.dth

Sample ID : CuSb 10mV

Creator method : Date :

Time:

Creator determ.: Date : 2000-08-14

Time: 17:08:32

Modified by : Date : 2017-07-13

Time: 13:43:34

Method : AB74_2b_Det of Cu Sb.mth

Title : Determination of Copper and Antimon. AB 74 part 2

Remark1 : 10 ml sample + 5 ml HCl (30%)

Remark2 :

Sample amount : 10.000 mL

Cell volume : 15.000 mL

Substance : Cu

Conc. : 4.180 ug/L

Conc.dev. : 0.139 ug/L (3.33%)

Amount : 62.696 ng

Add.amount : 50.000 ng

VR V nA I.mean Std.Dev. I.delta Comments

VR	V	nA	I.mean	Std.Dev.	I.delta	Comments
1 - 1	-0.271	1.763	1.773	0.014	0.000	
1 - 2	-0.275	1.783				
2 - 1	-0.275	3.125	3.097	0.039	1.324	
2 - 2	-0.275	3.070				
3 - 1	-0.271	4.506	4.529	0.032	1.431	
3 - 2	-0.275	4.551				

Substance : Sb(total)

Conc. : 1.577 ug/L

Conc.dev. : 0.042 ug/L (2.68%)

Amount : 23.648 ng

Add.amount : 50.000 ng

VR V nA I.mean Std.Dev. I.delta Comments

VR	V	nA	I.mean	Std.Dev.	I.delta	Comments
1 - 1	-0.160	3.94	3.96	0.038	0.00	
1 - 2	-0.160	3.99				
2 - 1	-0.160	12.47	12.35	0.178	8.38	
2 - 2	-0.160	12.22				
3 - 1	-0.156	20.50	20.59	0.135	8.25	
3 - 2	-0.160	20.69				

Substance Calibr. Y.reg/offset Slope Mean deviat. Corr.Coeff.

Substance	Calibr.	Y.reg/offset	Slope	Mean deviat.	Corr.Coeff.
Cu	std.add.	1.754e-009	4.196e-004	6.064e-011	0.99954
Sb(total)	std.add.	3.974e-009	2.521e-003	1.641e-010	0.99990

Final results +/- Res. dev. % Comments

Cu: default = 6.270 ug/l 0.209 3.334

Sb(total): default = 2.365 ug/l 0.063 2.682

Method print for the determination of Sb and Cu according to method 2

Method parameters

Method : AB074_2b_Det of Cu and Sb or Bi.mth

Title : Determination of Copper and Antimon or Bismuth. AB 74 Method 2

Remark1 : 10 ml sample + 5 ml HCl (30%)

Remark2 :

Calibration : Standard addition

Technique : Batch

Addition : Manual

Sample ID : Sample

Sample amount (mL): 10.000

Cell volume (mL): 15.000

Voltammetric parameters

Mode : DP - Differential Pulse

Highest current range : 10 mA
Lowest current range : 100 nA

Electrode : HMDE
Drop size (1..9) : 4
Stirrer speed (rpm) : 2000

Initial electr. conditioning : No

No. of additions : 2
No. of replications : 2

Measure blank : No
Addition purge time (s) : 20

Initial purge time (s) : 300

Conditioning cycles

Start potential (V)	:	0.000
End potential (V)	:	0.000
No. of cycles	:	0

Hydrodynamic (measurement)

Cleaning potential (V)	:	No
Cleaning time (s)	:	0.000
Deposition potential (V)	:	-0.400
Deposition time (s)	:	180.000

Sweep

Equilibration time (s)	:	10.000
Start potential (V)	:	-0.450
End potential (V)	:	-0.070
Voltage step (V)	:	0.004
Voltage step time (s)	:	0.200
Sweep rate (V/s)	:	0.020
Pulse amplitude (V)	:	0.010
Pulse time (s)	:	0.040

Cell off after measurement : Yes

Peak evaluation

Regression technique	:	Linear Regression
Peak evaluation	:	Height
Minimum peak width (V.steps)	:	10
Minimum peak height (A)	:	1.000e-010
Reverse peaks	:	No
Smooth factor	:	4
Eliminate spikes	:	Yes

Substances

Cu	:	-0.270 V	+/- 0.050 V
----	---	----------	-------------

Standard solution : 1 1.000 mg/L
Addition volume (mL) : 0.050

Copper : Final result (Cu) =
Conc * (15 / 10) * (1e+006 / 1) + 0 - 0

Sb/Bi : -0.160 V +/- 0.050 V

Standard solution : 2 1.000 mg/L
Addition volume (mL) : 0.050

Antimony/Bismuth : Final result (Sb/Bi) =
Conc * (15 / 10) * (1e+006 / 1) + 0 - 0

Baseline

Substance	Addition	automatic	start (V)	end (V)	type	scope
Cu	Sample	yes	---	---	linear	wholePeak
	Addition 1	yes	---	---	linear	wholePeak
	Addition 2	yes	---	---	linear	wholePeak
Sb/Bi	Sample	yes	---	---	linear	wholePeak
	Addition 1	yes	---	---	linear	wholePeak
	Addition 2	yes	---	---	linear	wholePeak

Report for the example determination of Sb, Bi and Cu in tap water according to method 3

===== METROHM 797 VA COMPUTRACE (Version 1.0.0.1) (Serial No. 0) =====

Determination : 08161516_SbBi tap water.dth

Sample ID : SbBi tap water

Creator method : Date : Time:

Creator determ.: Date : 2000-08-16 Time: 15:16:45

Modified by : Date : 2017-07-13 Time: 13:36:30

Method : AB74_3_Det of Sb Bi.mth

Title : Determination of Antimony and Bismuth. AB 74 part 2

Remark1 : 10 ml water + 0.6 ml HCl (30%)

Remark2 :

Sample amount : 10.000 mL

Cell volume : 10.600 mL

Substance : Sb(III)

Conc. : 2.828 ug/L

Conc.dev. : 0.073 ug/L (2.57%)

Amount : 29.981 ng

Add.amount : 50.000 ng

VR V nA I.mean Std.Dev. I.delta Comments

VR	V	nA	I.mean	Std.Dev.	I.delta	Comments
1 - 1	-0.117	8.22	8.20	0.031	0.00	
1 - 2	-0.117	8.18				
2 - 1	-0.114	22.31	22.09	0.311	13.89	
2 - 2	-0.114	21.87				
3 - 1	-0.114	35.13	35.07	0.080	12.98	
3 - 2	-0.114	35.02				

Substance : Bi

Conc. : 1.692 ug/L

Conc.dev. : 0.052 ug/L (3.09%)

Amount : 17.935 ng

Add.amount : 50.000 ng

VR V nA I.mean Std.Dev. I.delta Comments

VR	V	nA	I.mean	Std.Dev.	I.delta	Comments
1 - 1	-0.010	3.78	3.76	0.031	0.00	
1 - 2	-0.010	3.74				
2 - 1	-0.010	14.53	14.47	0.083	10.71	
2 - 2	-0.010	14.41				
3 - 1	-0.010	24.57	24.66	0.131	10.19	
3 - 2	-0.010	24.75				

Substance Calibr. Y.reg/offset Slope Mean deviat. Corr.Coeff.

Substance	Calibr.	Y.reg/offset	Slope	Mean deviat.	Corr.Coeff.
Sb(III)	std.add.	8.249e-009	2.917e-003	3.422e-010	0.99979
Bi	std.add.	3.814e-009	2.254e-003	1.960e-010	0.99991

Final results +/- Res. dev. % Comments

Sb(III): default = 2.998 ug/l 0.077 2.572

Bi: default = 1.794 ug/l 0.055 3.093

===== METROHM 797 VA COMPUTRACE (Version 1.0.0.1) (Serial No. 0) =====

Determination : 08161556_Cu tap water.dth

Sample ID : Cu tap water

Creator method : Date : Time:

Creator determ.: Date : 2000-08-16 Time: 15:56:29

Modified by : --- Date : Time:

Method : AB74_3_Det of Cu.mth

Title : Determination of Copper and Antimon. AB 74 part 2

Remark1 : 10.8 ml solution Sb Bi determination + 5 ml HCl (30%)

Remark2 : contains 10 ml sample

Sample amount : 10.000 mL

Cell volume : 15.800 mL

Substance : Cu

Conc. : 4.541 ug/L

Conc.dev. : 0.154 ug/L (3.39%)

Amount : 68.118 ng

Add.amount : 50.000 ng

VR	V	nA	I.mean	Std.Dev.	I.delta	Comments
1 - 1	-0.279	2.219	2.163	0.080	0.000	
1 - 2	-0.279	2.106				
2 - 1	-0.279	3.727	3.705	0.031	1.542	
2 - 2	-0.279	3.683				
3 - 1	-0.275	5.322	5.290	0.046	1.585	
3 - 2	-0.275	5.257				

Substance	Calibr.	Y.reg/offset	Slope	Mean deviat.	Corr.Coeff.
Cu	std.add.	2.154e-009	4.743e-004	7.231e-011	0.99948

Final results	+/- Res. dev.	%	Comments
Cu: default	= 6.812 ug/l	0.231	3.387

Method print for a determination of Sb, Bi and Cu according to method 3

```
===== METROHM 746 VA TRACE ANALYZER (5.746.0101) =====
Method: AB74_2c.mth          OPERATION SEQUENCE
Title : Determination of Sb, Bi and Cu. AB 74 Method 2c
```

Instructions	t/s	Main parameters	Auxiliary parameters	
1	SMP>M	V.fraction	mL	V.total L
2	DOS>M	Soln.name	HCl	V.add 0.600 mL
3	PURGE			
4	STIR	300.0	Rot.speed	2000 /min
5	(ADD			
6	PURGE			
7	STIR	20.0	Rot.speed	2000 /min
8	OPURGE			
9	(REP			
10	SEGMENT	Segm.name	SbBi_ASV	
11	REP)1			
12	PURGE			
13	ADD>M	Soln.name	Sb_Std	V.add 0.050 mL
14	ADD>M	Soln.name	Bi_Std	V.add 0.050 mL
15	ADD)2			
16	DOS>M	Soln.name	HCl	V.add 5.000 mL
17	PURGE			
18	STIR	300.0	Rot.speed	2000 /min
19	(ADD			
20	PURGE			
21	STIR	20.0	Rot.speed	2000 /min
22	OPURGE			
23	(REP			
24	SEGMENT	Segm.name	Cu_ASV	
25	REP)1			
26	PURGE			
27	ADD>M	Soln.name	Cu_Std	V.add 0.050 mL
28	ADD)2			
29	END			

Method: AB74_2c	SEGMENT	
	SbBi_ASV	

Instructions	t/s	Main parameters	Auxiliary parameters	
1	STIR	5.0	Rot.speed	2000 /min
2	HMDE		Drop size	4
3	DPMODE		U.ampl	10 mV
			t.step	0.20 s
4	MEAS	180.0	U.meas	-240 mV
5	MEAS	20.0	U.meas	-150 mV
6	0STIR	10.0		
7	SWEEP	18.2	U.start	-300 mV
			U.end	50 mV
8	0MEAS		U.standby	mV
9	END			

Method: AB74_2c	SEGMENT	
	Cu_ASV	

	Instructions	t/s	Main parameters	Auxiliary parameters
1	STIR	5.0	Rot.speed 2000 /min	
2	HMDE		Drop size 4	Meas.cell normal
3	DPMODE		U.ampl 10 mV	t.meas 20.0 ms
			t.step 0.20 s	t.pulse 40.0 ms
4	MEAS	180.0	U.meas -400 mV	
5	0STIR	10.0		U.step 4 mV
6	SWEEP	19.6	U.start -450 mV	Sweep rate 20 mV/s
			U.end -70 mV	
7	0MEAS		U.standby mV	
8	END			