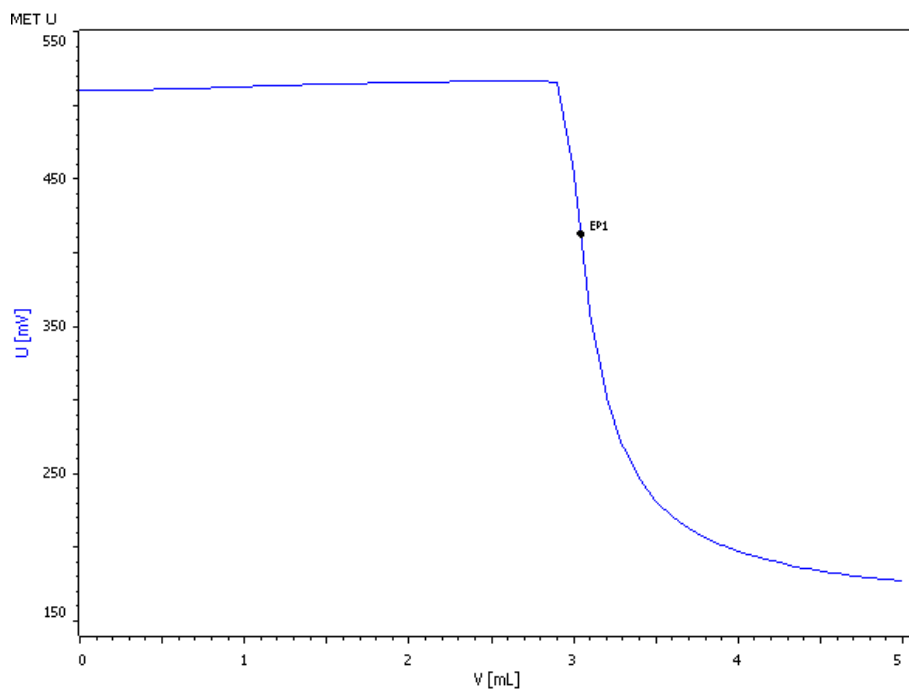


# Automated photometric determination of palladium using the Optrode



Palladium is determined at pH 4 to 5 by back-titration with zinc sulfate. To visualize the endpoint xylenol orange is used as indicator and the equivalence point is detected with the Optrode at a wavelength of 610 nm.

# Method description

## Sample

Aqueous solution of palladium (0.05 mol/L)

Xylenol orange indicator	100 mg xylenol orange disodium salt is dissolved in 100 mL deion. water.
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## Sample preparation

No sample preparation is required.

## Analysis

2 mL sample solution is pipetted into a 200 mL plastic beaker and 90 mL deion. water is added. After the addition of 10 mL acetate buffer, 10 mL  $c(\text{EDTA}) = 0.1 \text{ mol/L}$  and 10 mL xylenol orange indicator solution, the solution is titrated with  $c(\text{ZnSO}_4) = 0.1 \text{ mol/L}$  until after the equivalence point.

## Configuration

907 Titrand	2.907.0020
815 Robotic USB Sample Processor XL	2.815.0020
786 Swing head	2.786.0040
Swing arm	6.1462.070
Titration head	6.1458.010
Sample rack 28 x 200 mL	6.2041.830
800 Dosino, 5x	2.800.0010
802 Stirrer	2.802.0020
5 mL Dosing unit	6.3032.150
10 mL Dosing unit, 2x	6.3032.210
20 mL Dosing unit	6.3032.220
50 mL Dosing unit	6.3032.250
Disposable PP sample beaker, 200 mL	6.1459.310
Optrode	6.1115.000

## Parameters

Mode	MET U
Stirring rate	8
Pause	60 s
Signal drift	50 mV/min
Min. waiting time	5 s
Max. waiting time	26 s
Volume increment	0.1 mL
EP criterion	15 mV
EP recognition	Greatest

## Solutions

Titrant	$c(\text{ZnSO}_4) = 0.1 \text{ mol/L}$ 28.9 g $\text{ZnSO}_4 \cdot 7 \text{ H}_2\text{O}$ is weighted into a 1000 mL volumetric flask and dissolved in approx. 500 mL deion. water. After the addition of 0.5 mL $w(\text{H}_2\text{SO}_4) = 25\%$ , the solution is filled up to the mark with deion. water.
EDTA solution	$c(\text{EDTA}) = 0.1 \text{ mol/L}$ If possible this solution should be bought from a supplier.
Buffer solution pH = 4.9	123 g sodium acetate and 50 mL glacial acetic acid are given into a 1 L volumetric flask, dissolved in approx. 800 mL deion. water and the flask is filled up to the mark with deion. water.

## Results

Mean results (n = 6)

Pd content / (g/L)	10.400
s(rel) / %	1.39

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