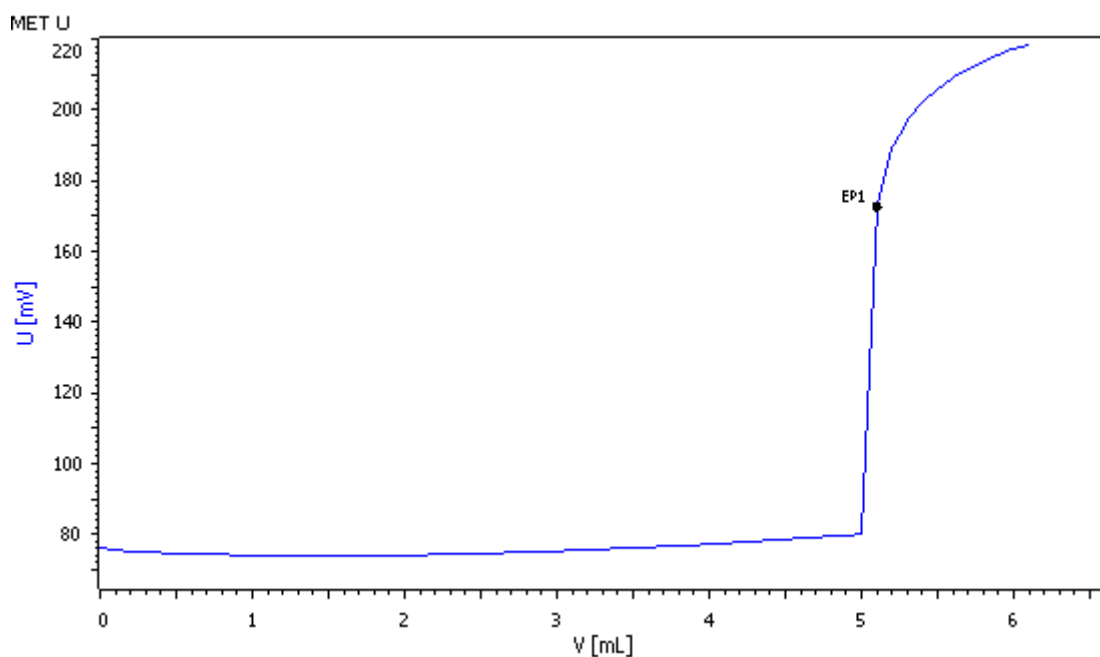


Titration Application Note T-105

Fully automated determination of bismuth(III) in aqueous solution



This Application Note describes the fully automated complexometric determination of bismuth(III) in aqueous solutions with a copper ion-selective electrode and the MATi 07 system.

Method description

Sample

Aqueous bismuth(III) solution

EP recognition greatest

Sample preparation

No sample preparation is necessary

Results

Mean in g/L	RSD in %
20.48 (n = 7)	1.44

Configuration

MATi 07

Ion-selective electrode, Cu 6.0502.140

LL ISE Reference 6.0750.100

Solutions

Titrant $c(\text{CuSO}_4) = 0.1 \text{ mol/L}$ in H_2O
If possible, this solution should be bought from a supplier

EDTA solution $c(\text{Na}_2\text{EDTA}) = 0.1 \text{ mol/L}$ in H_2O
If possible, this solution should be bought from a supplier

Acetate buffer 123 g Sodium acetate and 86 mL glacial acetic acid are dissolved in distilled water and filled up to 1 L

Analysis

The sample solution is diluted with approximately 50 mL distilled water in a titration beaker. Then, add 5 mL buffer solution and an excess (e.g., 10.0 mL) of EDTA solution. The excess of EDTA is then back-titrated with titrant $c(\text{CuSO}_4) = 0.1 \text{ mol/L}$ in H_2O .

Parameters

Mode MET U

Pause 30 s

Stirrer speed 8

Volume. increment 100 μL

Signal drift 50 mV/min

Max. waiting time 26 s

Stop EP 1

EP criterion 5 mV

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