

Ti Application Note No. T- 40

Title:	Determination of the phenylglycine content
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Summary:	Determination of phenylglycine by non-aqueous, potentiometric titration with sodium methylate using a special combined glass electrode.
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Sample:	Two different phenylglycine samples
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Sample Preparation:	none
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Instruments and Accessories:	702, 716, 736 or 751 Titrino or 726 Titroprocessor, 6.0239.100 combined glass electrode
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Analysis:	Weigh exactly ca. 300 mg sample into a beaker, then add 4 mL methylene chloride and 10 mL 3-chloroaniline. Stir the solution thoroughly for 4 min to dissolve the sample, add 50 mL chlorobenzene and titrate with $c(\text{CH}_3\text{ONa}) = 0.2 \text{ mol/L}$ using the MET mode.
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Calculation:	<p>1 mL $c(\text{CH}_3\text{ONa}) = 0.2 \text{ mol/L}$ corresponds to 30.234 mg phenylglycine.</p> <p>$\% \text{ phenylglycine} = \text{EP1} * \text{C01} * \text{C02} / \text{C00}$</p> <p>EP1 = titrant consumption in mL C00 = ca. 300 (sample mass in mg) C01 = 30.234 C02 = 100 (conversion factor for %)</p>
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Remarks:	<ul style="list-style-type: none">– Stir the solution thoroughly during the titration.– Use alternately two different electrodes for the titrations. After three titrations replace the electrode used by the new one and immerse it in dist. water for 5 min before using it again.– Rinse the electrode after each titration first with ethanol and then with dist. water. <p>Results:</p> <p>Sample A: $\text{AVG}(3) = 90.47 \pm 0.26 \%$ phenylglycine</p> <p>Sample B: $\text{AVG}(3) = 87.37 \pm 0.24 \%$ phenylglycine</p>
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