

Ti Application Note No. T- 27

Title: Alkalinity of amine-containing gas washing solutions

Summary: Determination of the alkalinity of gas washing solutions containing alkanolamines by potentiometric titration with sulphuric acid using the combined glass electrode.

Sample: Gas washing solutions containing alkanolamines, e.g. monoethanolamine (MEA), diglycolamine (DGA) or diisopropanolamine ADIP)

Sample Preparation: none

Instruments and Accessories: 702, 716 or 736 Titrino or 726 Titroprocessor, 6.0219.100 combined glass electrode

Analysis: Pour ca. 50 mL dist. water into a beaker, add 2 ... 3 g sample solution with a syringe (the exact sample mass is determined by backweighing) and titrate with $c(\text{H}_2\text{SO}_4) = 0.05 \text{ mol/L}$ (0.1 N) to an endpoint at $\text{pH} = 4.8$.

Calculation:

$$\% \text{ MEA} = \text{EP1} * \text{C01} / \text{C00}$$
$$\% \text{ DGA} = \text{EP1} * \text{C02} / \text{C00}$$
$$\% \text{ ADIP} = \text{EP1} * \text{C03} / \text{C00}$$

EP1 = titrant consumption in mL
C00 = sample weight in g
C01 = 0.61
C02 = 1.05
C03 = 1.33

Remarks: As the samples can also contain sulphides you have to use a special combined glass electrode whose bridge electrolyte has to be changed every day.